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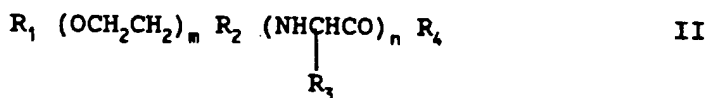
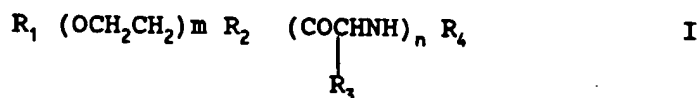
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(54) **Physical trapping type polymeric micelle drug preparation.**

(57) A polymeric micelle type drug comprises at least one hydrophobic drug physically trapped in a drug carrier comprising a block copolymer represented by formula I or II:



wherein R_1 is hydrogen or an alkyl group having 1 to 20 carbon atoms, R_2 is NH, CO, $R_6(CH_2)_qR_7$ (in which R_6 represents OCO, OCONH, NHCO, NHCOO, NHCONH, CONH or COO, R_7 represents NH or CO, and q is 1 to 6), R_3 is hydrogen, an alkyl group having 1 to 20 carbon atoms, $(CH_2)_pC_6H_5$, $(CH_2)_pCOOR_5$ or CH_2CONHR_5 (in which p is 1 or 2, R_5 represents a C_{1-20} alkyl group, a benzyl-substituted C_{1-20} alkyl group or a benzyl group), R_4 is hydrogen, hydroxyl, RCO-, RNH- or RO- where R is an alkyl group having 1 to 20 carbon atoms, m is 4-2500, and n is 2-300.

FIELD OF THE INVENTION

The present invention relates to drug carriers having hydrophilic and hydrophobic segments capable of physically trapping hydrophobic drugs, as well as to a polymeric micelle type drug having hydrophilic drugs physically trapped to said carrier.

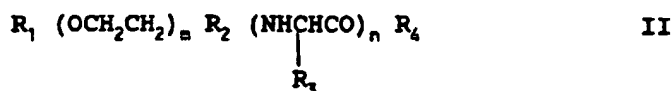
BACKGROUND OF THE INVENTION

A polymeric micelle type drug, in which a hydrophobic drug is chemically bound to a block copolymer through a covalent bond, was successfully constructed and applied by the present inventors for a patent in Japanese Patent Application No. 116,082/89. In spite of the fact that this prior polymeric micelle type drug is extremely superior as the means of administering a hydrophobic drug, the combination of hydrophobic drug and block copolymer is disadvantageously limited because its preparation requires functional groups for chemically binding a hydrophobic drug to a block copolymer.

Under the circumstances, however, no development has been made in a method of physically trapping hydrophobic drugs so as to incorporate them in the inner core of polymeric micelle or in a drug carrier for such a method.

The present inventors have tried to develop a physical trapping type polymeric micelle drug, in order to solve the above disadvantage of the chemical bond type polymeric micelle drug. The present inventors, as a result of their eager research, succeeded in preparing polymeric micelle type drug applicable to a wide variety of combinations of hydrophobic drugs and block copolymer by constructing a polymeric micelle from a drug carrier composed of a block copolymer having hydrophilic and hydrophobic segments and then permitting hydrophobic drugs to be physically trapped into the hydrophobic inner core of said micelle. The system for trapping drugs, developed by the present inventors, allows a wide variety of hydrophobic drugs to be easily incorporated in the polymeric micelle.

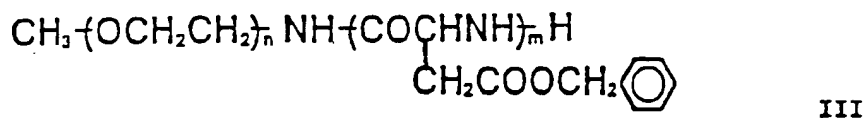
The present invention comprises a polymeric micelle type drug composition which comprises at least one hydrophobic drug physically trapped in a drug carrier comprising a block copolymer represented by formula I or II;



wherein R_1 is hydrogen or an alkyl group having 1 to 20 carbon atoms, R_2 is NH, CO, $R_6(CH_2)_qR_7$ (in which R_6 represents OCO, OCONH, NHCO, NHCOO, NHCONH, CONH or COO, R_7 represents NH or CO, and q is 1 to 6), R_3 is hydrogen, an alkyl group having 1 to 20 carbon atoms, $(CH_2)_pC_6H_5$, $(CH_2)_pCOOR_5$ or CH_2CONHR_5 (in which p is 1 or 2, R_5 represents a C_{1-20} alkyl group, a benzyl-substituted C_{1-20} alkyl group or a benzyl group), R_4 is hydrogen, hydroxyl, RCO-, RNH- or RO- where R is an alkyl group having 1 to 20 carbon atoms, m is 4-2500, and n is 2-300.

An alkyl group has from 1 to 20 carbon atoms and may be either straight chain or branched. Preferably the alkyl group has from 1 to 10 carbon atoms, for example 1, 2, 3, 4 or 5 carbon atoms.

According to a preferred embodiment the block copolymer is a compound represented by formula III:



wherein m is 4-2500 and n is 2-300.

According to a further preferred embodiment, the hydrophobic drug is adriamycin or indomethacin.

The present invention further provides a method for trapping hydrophobic drugs in drug carrier, which comprises the heating, ultrasonication or organic solvent treatment of hydrophobic drugs and drug carrier or formula I, II or III to physically trap said hydrophobic drugs in polymeric micelles comprising said drug carrier.

FIG. 1 shows particle size distribution by dynamic light scattering of polymeric micelles of polyethylene oxide-poly-(β -benzyl L-aspartate) block copolymer (A-5-10) in an aqueous solution.

FIG. 2 shows changes in fluorescence spectra of pyrene when incorporated in the inner core of polymeric micelle by heating. In the figure, Nos. 1-8 indicate the fluorescence spectra of pyrene at the respective concentrations of block copolymer.

FIG. 3 shows the amount of pyrene incorporated by three methods at various concentrations of block copolymer.

FIG. 4 is a gel permeation chromatogram (GPC) of adriamycin incorporated into polymeric micelles.

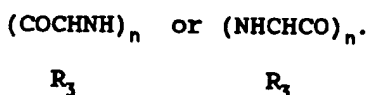
FIG. 5 is a gel permeation chromatogram of polymeric micelles.

FIG. 6 is a gel permeation chromatogram of adriamycin incorporated in micelles after allowed to stand for 5 hours in the presence of 50%(V/V) of fetal bovine serum.

FIG. 7 is a gel permeation chromatogram of 50%(V/V) fetal bovine serum.

Fig. 8 is a gel permeation chromatogram monitored at 312 nm at which indomethacin shows characteristic absorption.

The block copolymer comprises a hydrophilic segment and a hydrophobic segment. The hydrophilic segment comprises polyethylene oxide and the hydrophobic segment is represented by



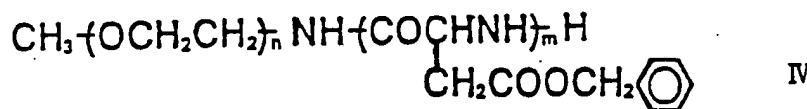
The hydrophobic segment comprises, for example, poly(β -benzyl L-aspartate), poly(γ -benzyl L-glutamate), poly(β -substituted aspartate), poly(γ -substituted glutamate), poly(L-leucine), poly(L-valine) or poly(L-phenylalanine).

The drug to be physically trapped in the hydrophobic inner core of polymeric micelle is not particularly limited. Examples are anticancer drugs such as adriamycin, daunomycin, methotrexate, mitomycin C, etc., painkilling and anti-inflammatory drugs such as indomethacin etc., drugs for the central nervous system, drugs for the peripheral nervous system, drugs against allergies, drugs for the circulatory organs, drugs for the respiratory organs, drugs for the digestive organs, hormones as drugs, metabolizing drugs, antibiotics, drugs for use in chemotherapy, etc.

The physical means of trapping hydrophobic drugs in polymeric micelles composed of the present drug carrier includes heating, ultrasonication and organic solvent treatment, which are conducted solely or in combination with one another. Heating is carried out at 30-100 °C for a period of time from 10 min. to 24 hours. Ultrasonication is carried out in the range of 1-200 W for a period of time from 1 second to 2 hours. The organic solvent used in organic solvent treatment is DMF, DMSO, dioxane, chloroform, n-hexane, toluene, methylene chloride, etc., which is used in the absence of water or after added in an amount of 0.01 % (v/v) or more to water.

Hereinafter, the present invention is specifically explained in detail with reference to the actual incorporation of adriamycin as an anticancer drug, indomethacin as a painkilling, anti-inflammatory drug and pyrene as a typical hydrophobic chemical, into an AB type block copolymer composed of a hydrophilic segment derived from a derivative of polyethylene oxide and a hydrophobic segment of poly(β -benzyl L-aspartate).

The compound of formula IV:



is polyethylene oxide-poly(β -benzyl L-aspartate) block copolymer consisting of polyethylene oxide and poly(β -benzyl L-aspartate) which have hydrophilic and hydrophobic properties, respectively. The compound of formula IV is compound of formula I wherein R₁ is a methyl group, R₂ is NH, R₃ is CH₂COOCH₂C₆H₅, and R₄ is

H.

This block copolymer is prepared by polymerizing, in the presence of an initiator, β -benzyl L-aspartate N-carboxy anhydride from the terminal primary amino group of polyethylene oxide (molecular weight of 200-250,000) having an amino group in one terminal and a methoxy group at the other terminal. The portion of poly(β -benzyl L-aspartate) in the block copolymer polyethylene oxide-poly(β -benzyl L-aspartate) may have a molecular weight varying from 205 to 62,000. By suitable selection of a chain length ratio of the two segments, this block copolymer forms a polymeric micelle with ethylene oxide as an outer shell and poly(β -benzyl L-aspartate) as an inner core. This polymeric micelle can stably incorporate hydrophobic pyrene, adriamycin and indomethacin by heating, ultrasonication, or treatment with organic solvent.

The drug carrier composed of the block copolymer according to the invention forms a stable polymeric micelle structure with which hydrophobic drugs can be incorporated very efficiently via physical trapping into the inner core. A drug difficult to administer into the living body owing to sparing water-solubility for its high hydrophobicity can be administered in the form of polymeric micelle type drug.

In addition, the invention do not require any functional group for chemical bonding and thereby enables a wide variety of combinations of hydrophobic drugs and polymeric micelle.

EXAMPLES

The present invention is described in detail with reference to the following examples, which however are not intended to limit the scope of the invention.

Example 1

β -benzyl L-aspartate N-carboxylic anhydride (1.99 g) was dissolved in 3 ml N,N-dimethylformamide, followed by addition of 15 ml of chloroform. Separately, 4.00 g of polyethylene oxide having methoxy group in one terminal and an amino group in the other terminal (molecular weight: 5,000) was dissolved in 15 ml chloroform, and the solution was then added to the above solution of β -benzyl L-aspartate N-carboxy anhydride. 26 hours thereafter, the reaction mixture was added dropwise to 330 ml diethyl ether, thereby giving rise to polymer precipitates which in turn were recovered by filtration, then washed with diethyl ether and dried under vacuum, to give polyethylene oxide poly(β -benzyl L-aspartate) block copolymer (referred to as "PEO-PBLA," hereinafter) (A-5-10). Yield was 5.13 g (91 %). The compositions of block copolymers thus synthesized are summarized in Table 1.

Table 1

Characterization of Polyethylene Oxide-Poly(β -Benzyl L-Aspartate) Block Copolymer and Micelles

Sample	PEO wt (%) ^a	Mn ^a	nPEO	nPBLA ^a	Particle size (nm) ^b	CMC (mg/L)
A-5-10	73.0	7000	110	9.0	18	10
A-5-20	53.3	9100	110	19	17	5.0
A-12-20	35.0	16000	270	20	21	10

a) determined by ¹H-NMR

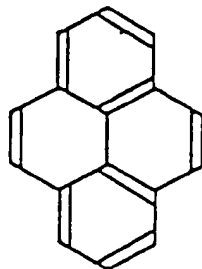
b) determined by dynamic light scattering (number-average)

Example 2 Formation of Micelles

The block copolymer synthesized in Example 1 was dissolved at a concentration of 0.01-0.1 % (w/v) in water or a suitable buffer. The formation of micelles in the thus obtained solutions was ascertained by measurement of distribution of particle size by dynamic light scattering. The result is set forth in FIG. 1. The particle size of micelle and critical micelle concentration are also shown in Table 1.

Example 3 Incorporation of Pyrene into Micelles

Pyrene of formula V:



V

is sparingly soluble in water so that a predetermined amount of pyrene was dissolved at acetone. After dissolved, acetone was removed under a nitrogen atmosphere, and a micelle solution of PEO-PBLA (A-5-10) in distilled water was added at a concentration shown in Table 1 to the pyrene.

1. Incorporation by Stirring

The above mixture was stirred for 2 days so that pyrene was incorporated into micelles.

2. Incorporation by Heating

The above mixture was heated at 80 °C for 2 hours so that pyrene was incorporated into micelles.

3. Incorporation by Ultrasonication

The above mixture was ultrasonicated for 15 seconds so that pyrene was incorporated into micelles.

4. Incorporation by Treatment with DMF for Making the PBLA segment swelled in the PEO-PBLA micelle.

As described above, acetone was removed from the pyrene solution. To the pyrene was added DMF in an amount of 30 % relative to the micelle solution to be added afterward. Then, a solution of PEO-PBLA in distilled water was then added in a concentration shown in Table 3 to the pyrene solution. After stirred for 15 hours, the solution was dialyzed in a dialysis tube Spectrapor 6 (cut off molecular weight = 1,000) against water. According to the above procedure, pyrene was incorporated into micelles.

As is evident from increases in the intensities of the fluorescence spectra of the heated sample shown in FIG. 2, the incorporation of pyrene into micelles was confirmed in every incorporation means. FIG. 3 shows a comparison between the amounts of pyrene incorporated into micelles, where the incorporation means by heating attains the amount of incorporated pyrene as approx. 250 times high as the amount of pyrene saturated in water. Table 2 shows the partition coefficient of pyrene into PEO-PBLA (A-5-10) micelle relative to water.

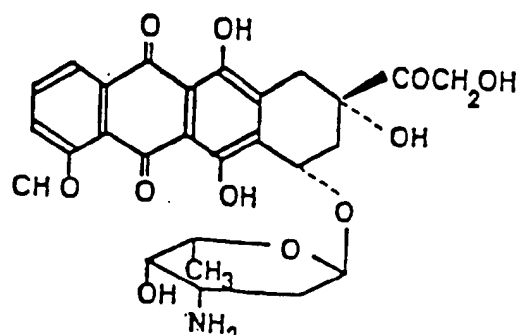
Table 2

Distribution Coefficient into Micelle Solution of Polyethylene Oxide-Poly(β -Benzyl L-Aspartate) Block Copolymer	
Means of Incorporating Pyrene	Distribution Coefficient (Kn)
Stirring	17000
Heating at 80 °C	21000
Ultrasonication	17000

Example 4

5 mg of adriamycin hydrochloride and 5 mg of PEO-PBLA (A-12-20) were added to 5 ml of 0.1 M Tris buffer, pH 9.1. Then, adriamycin was made miscible into micelles by stirring and ultrasonication.

Adriamycin is the compound of the following formula:



VI

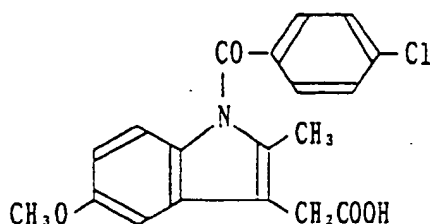
This compound itself does not dissolve in Tris buffer, pH 9.1, but can be completely dissolved according to the above procedure. As shown in FIG. 4, adriamycin appeared in gel-exclusion volume in GPC where the sample was monitored at 485 nm at which adriamycin shows characteristic absorption, and this indicates sufficient incorporation of adriamycin into micelles. In FIG. 4, elution volume is indicated as numerical values where 1.792, 3.292 and 9.300 mean micelles, a single polymer and unincorporated adriamycin, respectively.

Example 5

4.4 μ l triethylamine and 20 mg PEO-PBLA block copolymer (A-12-20) were added to a solution of 14 mg adriamycin hydrochloride in 4 ml DMF, and the mixture was stirred for 10 min. and then dialyzed for 15 hours against distilled water. Dynamic light scattering indicated that the sample thus obtained formed polymeric micelles with a weight-average diameter of 55 nm. FIG. 5 shows a gel permeation chromatogram of the polymeric micelles monitored at 485 nm. Adriamycin was incorporated in the micelles, as can be seen from its elution as micelles in gel exclusion volume (4.2-4.3 ml). FIG. 6 shows a gel permeation chromatogram of adriamycin incorporated in micelles after allowed to stand for 5 hours in the presence of 50%(V/V) fetal bovine serum. In FIG. 6, the peak (4.25 ml) eluted in gel exclusion volume and not present in the serum itself was not lowered in the presence of serum (Fig. 7), which indicates that adriamycin can be stably maintained in micells even in the presence of serum.

Example 6

15 mg of indomethacin of formula

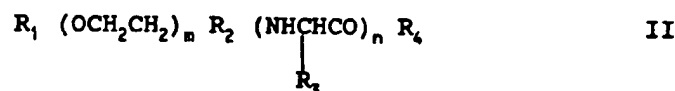


VII

as an anti-inflammatory drug was dissolved in 4 ml DMF, followed by addition of 20 mg of PEO-PBLA block copolymer (A-12-20). The mixture was stirred for 15 hours and dialyzed for 3 hours against 0.1 M phosphate buffer, pH 7.4, and then against water for 6 hours. The resulting sample was found to form polymeric micelles with a weight-average diameter of 56 nm, as determined by dynamic light scattering. FIG. 8 shows a gel permeation chromatogram monitored at 312 nm at which indomethacin shows characteristic absorption. The indomethacin was eluted as micelles in gel exclusion volume, indicating the incorporation of the indomethacin into micelles. 0.76 mg of indomethacin was found to be incorporated in the micelles from its adsorption monitored at 312 nm in a solvent of DMF/distilled water (7 : 3).

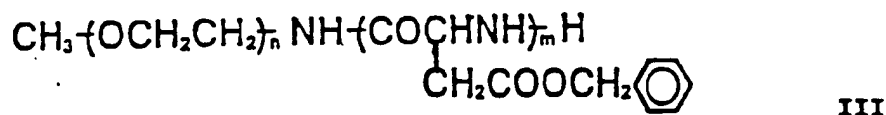
Claims

1. A polymeric micelle type drug composition which comprises at least one hydrophobic drug physically trapped in a drug carrier comprising a block copolymer represented by formula I or II:



wherein R_1 is hydrogen or an alkyl group having 1 to 20 carbon atoms, R_2 is NH, CO, $R_6(CH_2)_qR_7$ (in which R_6 represents OCO, OCONH, NHCO, NHCOO, NHCONH, CONH or COO, R_7 represents NH or CO, and q is 1 to 6), R_3 is hydrogen, an alkyl group having 1 to 20 carbon atoms, $(CH_2)_pC_6H_5$, $(CH_2)_pCOOR_5$ or CH_2CONHR_5 (in which p is 1 or 2, R_5 represents a C_{1-20} alkyl group, a benzyl-substituted C_{1-20} alkyl group or a benzyl group), R_4 is hydrogen, hydroxyl, RCO-, RNH- or RO- where R is an alkyl group having 1 to 20 carbon atoms, m is 4 to 2500, and n is 2 to 300.

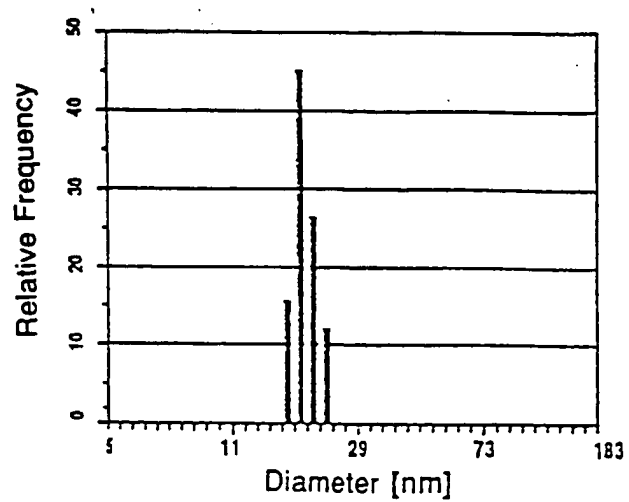
2. A composition according to claim 1, in which the block copolymer is a compound represented by formula III:



wherein m is 4 to 2500 and n is 2 to 300.

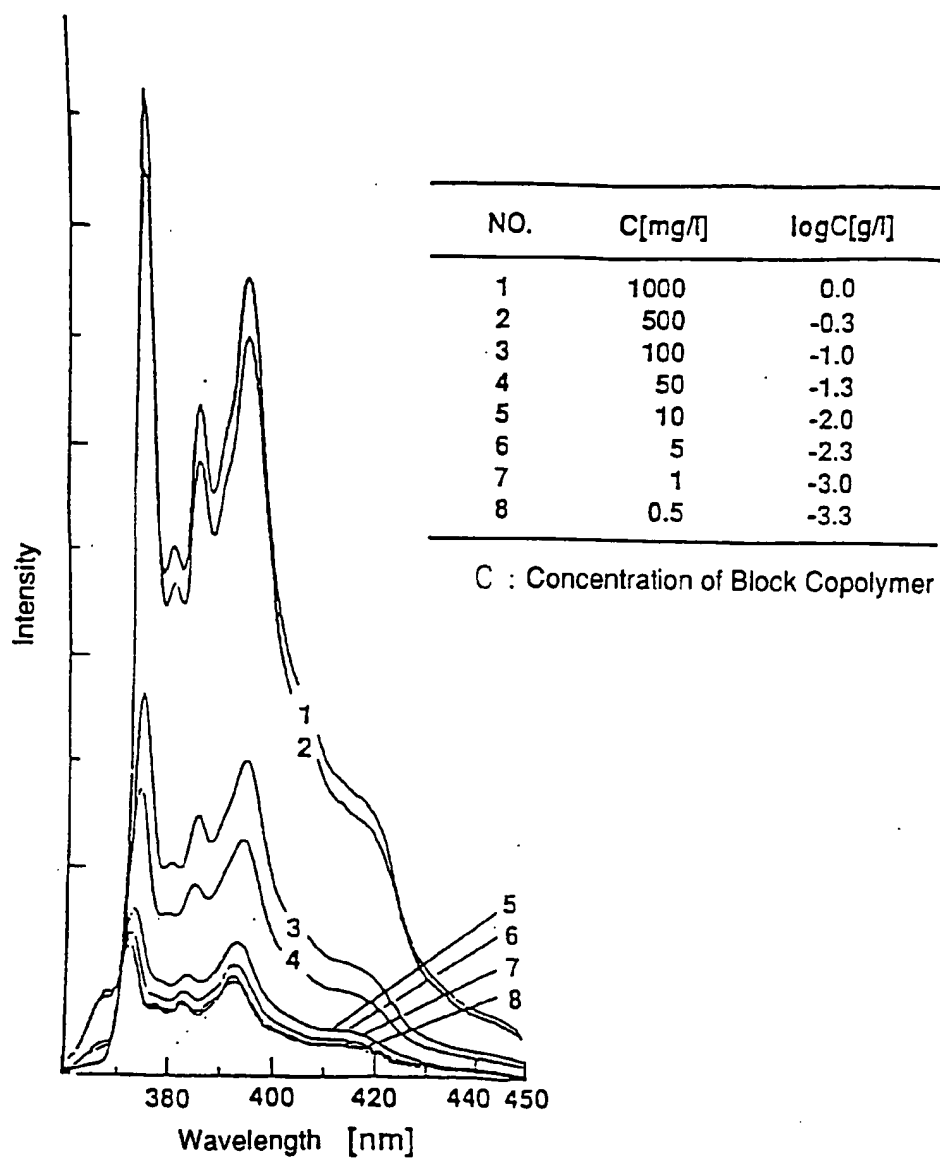
3. A composition according to claim 1 or 2, in which the hydrophobic drug is at least one of adriamycin, daunomycin, methotrexate, mitomycin C or indomethacin.
4. A composition according to claim 3 in which the hydrophobic drug is indomethacin or adriamycin.
5. A method for trapping hydrophobic drugs, which comprises the heating, ultrasonication or organic solvent treatment of hydrophobic drugs and drug carrier of formula I, II or III to physically trap said hydrophobic drugs in polymeric micelles comprising said drug carrier.

Fig. 1



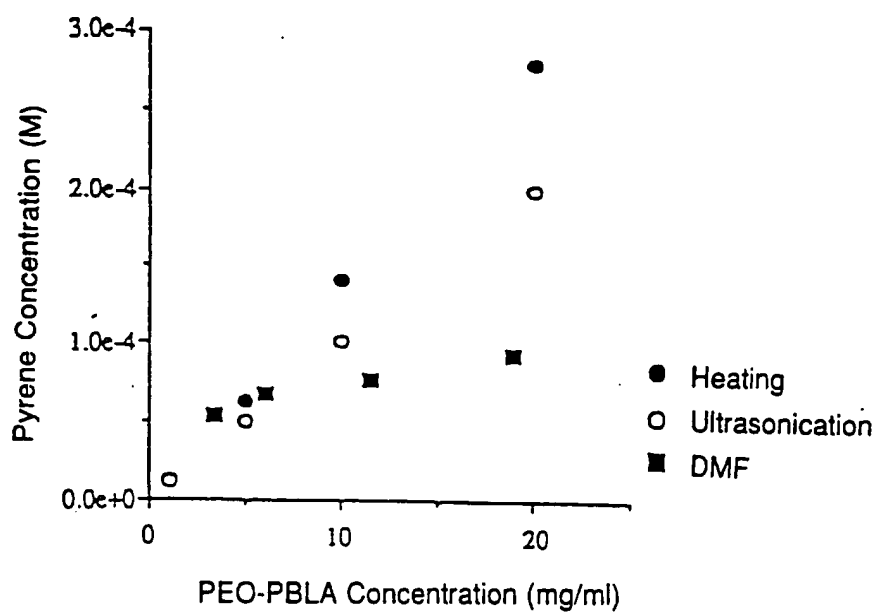
Particle Size Distribution of Block Copolymer
Micelles (A-5-10)

Fig. 2



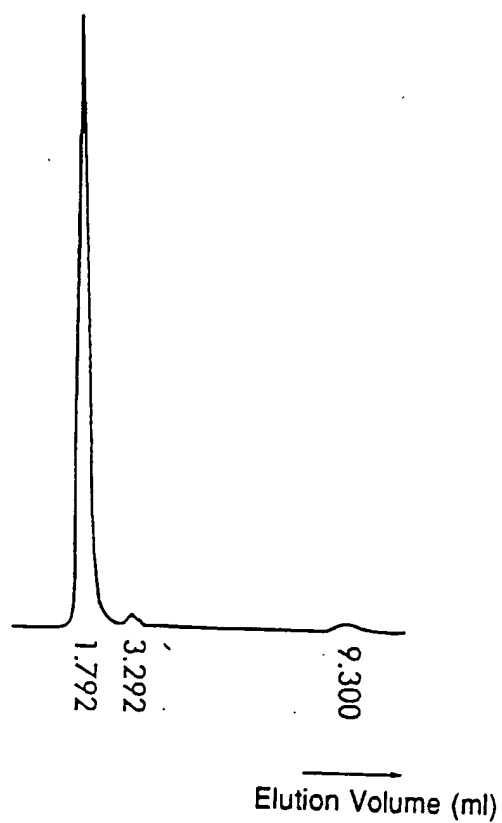
Fluorescence Spectra of Aqueous Pyrene Solution ($6 \times 10^{-7}M$)
 in the Presence of Block Copolymer Micelles(A-5-10)
 Excitation Wavelength: 339 nm

Fig. 3



Effect of Incorporation Means and Block Copolymer Concentration
on the Amount of Pyrene Incorporated into Block Copolymer Micelles
(A-5-10)

Fig. 4



Gel Permeation Chromatogram of Adriamycin-Incorporated Micelles

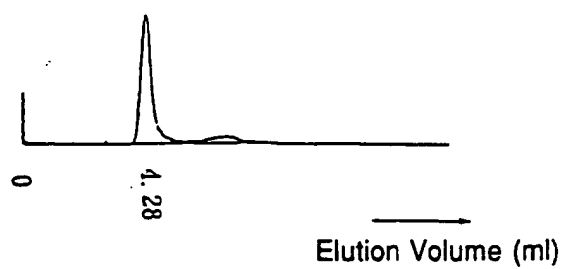
Column: Asahipak GS-510M

Eluent solvent: 0.1 M phosphate buffer, pH 7.4

Flow rate: 1.0 ml/min.

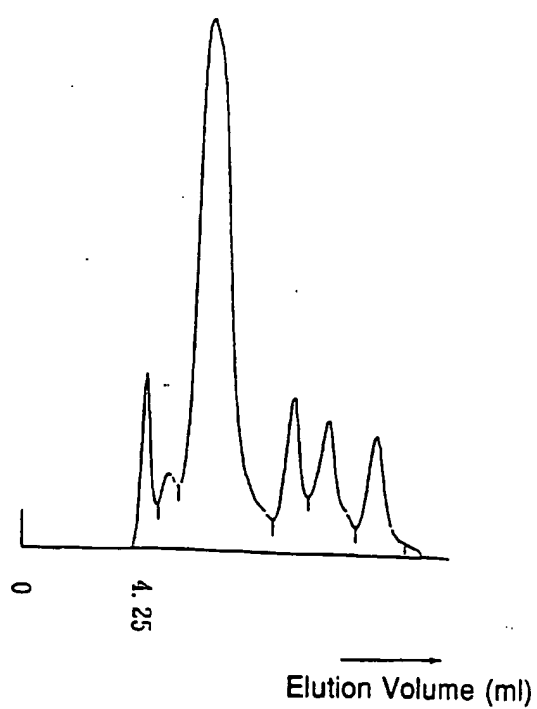
Adriamycin Concentration: 10 μ g/ml

Fig. 5



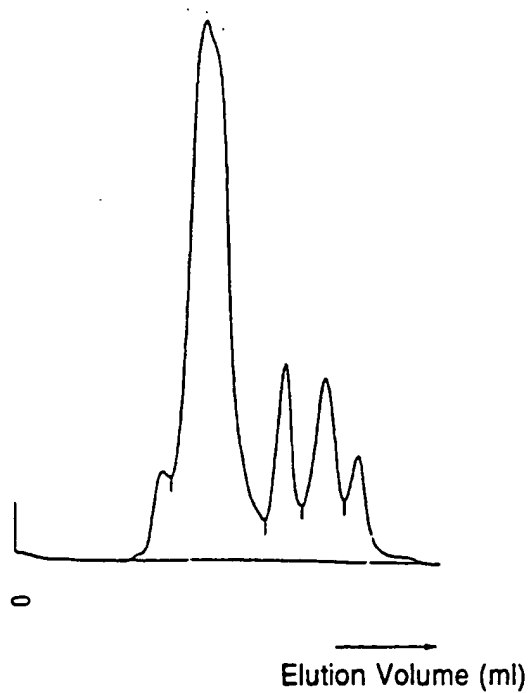
Column: Asahipak GS-520H
Eluent solvent: 0.1 M phosphate buffer, pH 7.4
Flow rate: 1.0 ml/min.
Detection: 485 nm

Fig. 6



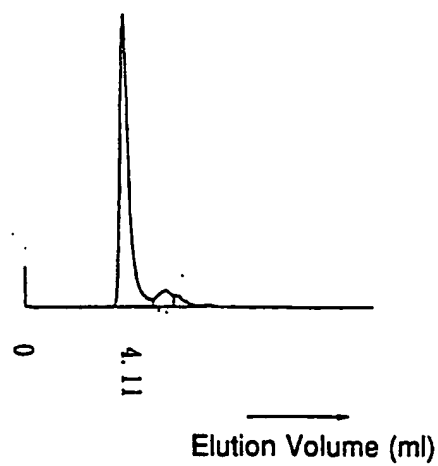
Column: Asahipak GS-520H
Eluent solvent: 0.1 M phosphate buffer, pH 7.4
Flow rate: 1.0 ml/min.
Detection: 485 nm

Fig. 7



Column: Asahipak GS-520H
Eluent solvent: 0.1 M phosphate buffer, pH 7.4
Flow rate: 1.0 ml/min.
Detection: 485 nm

Fig. 8



Column: Asahipak GS-520H
Eluent solvent: 0.1 M phosphate buffer, pH 7.4
Flow rate: 1.0 ml/min.
Detection: 312 nm



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(54) **Sustained release pharmaceutical composition.**

(57) A sustained release pharmaceutical pellet composition suitable for treating pain-associated conditions in patients comprises a core element including at least one active ingredient having an aqueous solubility of at least 1 in 30; and a core coating for the core element which is partially soluble at highly acidic pH and wherein the active ingredient, e.g. a morphine compound, is available for absorption at a relatively constant rate in the intestine over an extended period of time.

EP 0 609 961 A1

The present invention relates to a sustained release pharmaceutical composition, in particular a sustained release pharmaceutical composition including an active ingredient of high solubility in water, and to a method of preparing same.

As is known in the prior art, it is desirable in the treatment of a number of diseases, both therapeutically and prophylactically to provide the active pharmaceutical ingredient in a sustained release form. Desirably the sustained release provides a generally constant rate of release over an extended period. Whilst there is known in the prior art numerous sustained release formulations, the extension of sustained release regimens to active pharmaceutical ingredients of high solubility in water has been extremely limited. It has been found in the prior art that the high solubility in water of the active ingredient tends to generate a product which is susceptible to the phenomenon known as "dose dumping". That is, release of the active ingredient is delayed for a time but once release begins the rate of release is very high. Moreover, fluctuations tend to occur in the plasma concentrations of active ingredient which increases the likelihood of toxicity. Further, some degree of diurnal variation in plasma concentration of active ingredient has also been noted.

Prior art preparations may also suffer from other disadvantages, for example bioavailability of prior art preparations may be compromised by food. This is important since complex dosage regimens may lead to non-compliance.

For example, typical highly water soluble active ingredients include the opioid drugs which still play a major role in the treatment of acute and chronic pain, particularly pain associated with terminal diseases such as cancer.

Morphine is regarded as the opioid drug of choice in the treatment of cancer pain. It is universally acknowledged that the oral route of administration is preferred if sufficient pain relief can be obtained with an acceptable profile of side effects with respect to incidence and severity. Until recently, the liquid or immediate release tablet formulations of morphine were the only dosage forms available to physicians for oral administration in the treatment of cancer pain.

The oral administration of morphine has had many critics in the prior art who point to a supposed lack of efficacy. However, the accumulated evidence, particularly from the hospice environment, indicates that this criticism is unfounded if the dose and dosing interval are specifically optimised for each patient, the morphine doses are administered before the pain returns and in a strictly regular regimen. In practical terms, this means morphine doses ranging from 10 mg to in excess of 500 mg with dosing intervals ranging from every 2 to 6 hours. A relationship between blood morphine concentration and pain relief has been established in the treatment of post-operative and cancer pain.

The studies propose that there is a minimum effective concentration (MEC) for morphine for each patient. There is a five-fold interpatient variation in MEC in the treatment of post-operative pain and an even greater variation for cancer pain. This concept of a MEC for opioids has also been demonstrated for pethidine, methadone, fentanyl and ketobemidone. Repeated oral or parenteral doses produce fluctuating blood opioid concentrations, with the peak concentrations sometimes resulting in side effects, while the trough concentrations are usually associated with inadequate pain relief. Therefore, a formulation of morphine which reduces the fluctuations in blood opioid concentrations and has a longer duration of pain relief (e.g. a sustained release preparation) has widespread potential to improve pain relief in terminal care.

Currently, there is only one such preparation (MST Continus or MS Contin) being marketed world-wide. However, the combined pharmacokinetic and pharmacodynamic data suggest that this product is actually a delayed release formulation with some sustained release characteristics. While the manufacturers recommend a 12 hour dosing interval, extensive clinical experience suggests that an 8 hour interval is more realistic for continuous pain control.

Accordingly, it is an object of the present invention to overcome, or at least alleviate, one or more of the difficulties related to the prior art.

Accordingly, in a first aspect of the present invention there is provided a sustained release pharmaceutical pellet composition including

a core element including at least one active ingredient of high solubility; and
a core coating for the core element which is partially soluble at a highly acidic pH to provide a slow rate of release of active ingredient and wherein the active ingredient is available for absorption at a relatively constant faster rate in the intestine over an extended period of time, such that blood levels of active ingredient are maintained within the therapeutic range over an extended period of time.

By "sustained release" as used herein we mean release of active ingredient at such a rate that blood levels are maintained within the therapeutic range but below toxic levels over an extended period of time e.g. 10 to 24 hours or greater. By "active ingredient of high water solubility" as used herein we mean pharmaceutically active, orally acceptable ingredients having an aqueous solubility of approximately 1 in 30 or above.

By "bioavailability" as used herein we mean the extent to which the active drug ingredient is absorbed from the drug product and becomes available at the site of drug action.

The active ingredients of high solubility may be selected from antihistamines, antibiotics, antituberculosis agents, colinergic agents, antimuscarinics, sympathomimetics, sympatholytic agents, autonomic drugs, iron preparations, haemostatics, cardiac drugs, antihypertensive agents, vasodilators, non-steroidal antiinflammatory agents, opiate agonists, anticonvulsants, tranquilisers, stimulants, barbiturates, sedatives, expectorants, antiemetics, gastrointestinal drugs, heavy metal antagonists, antithyroid agents, genitourinary smooth muscle relaxants and vitamins. The invention is applicable to active ingredients of high solubility whether the solubility characteristics are pH dependent or pH independent.

Examples of active ingredients of high solubility are set out in the table below.

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	DRUG	SOLUBILITY (AQUEOUS)	pKa
	<u>Antihistamines</u>		
5	Azatadine maleate	very soluble	9.3
	Brompheniramine maleate	1 in 5	3.59, 9.12
	Carbinoxamine maleate	1 in 1	8.1
	Chlorpheniramine maleate	1 in 4	9.2
10	Dexchlorpheniramine maleate	1 in 1.1	
	Diphenhydramine HCl	1 in 1	9.0
	Doxylamine succinate	1 in 1	5.8, 9.3
	Methdilazine HCl	1 in 2	7.5
	Promethazine	1 in 0.6	9.1
15	Trimeprazine Tartrate	1 in 4	
	Tripelennamine citrate	1 in 1	3.9, 9.0
	Tripelennamine HCl	1 in 1	
20	Triprolidine HCl	1 in 2	3.6, 9.0
	<u>Antibiotics</u>		
	Penicillin V Potassium	1 in 1.5	0.5
	Cloxacillin sodium	1 in 2.5	2.7
25	Dicloxacillin sodium	freely soluble	2.7
	Nafcillin Sodium	freely soluble	2.7
	Oxacillin Sodium	1 in 3.5	2.8
	Carbenicillin Indanyl Sodium	freely soluble	2.6, 2.7, 3.3
	Oxytetracycline HCl	1 in 2	3.3, 7.3, 9.1
30	Tetracycline HCl	1 in 10	3.3, 7.7, 9.7
	Clindamycin Phosphate	1 in 2.5	7.7
	Clindamycin HCl	1 in 2	7.7
	Clindamycin Palmitate HCl	freely soluble	-
35	Lincomycin HCl	1 in 1	7.6
	Novobiocin Sodium	1 in 5	4.2, 9.1
	Nitrofurantoin Sodium	soluble	7.2
	Metronidazole hydrochloride	1 in 1	2.6
40	<u>Antituberculosis Agents</u>		
	Isoniazid	1 in 8	1.8, 3.5, 10.8
	<u>Cholinergic Agents</u>		
45	Amбенonium chloride	1 in 5	
	Bethanecol chloride	1 in 1	
	Neostigmine bromide	1 in 0.5	12.0
	Pyridostigmine bromide	1 in 1	
	<u>Antimuscarinics</u>		
50	Anisotropine methylbromide	soluble	

	Clidinium bromide	soluble	
	Dicyclomine HCl	1 in 20	9
	Glycopyrrolate	1 in 5	
5	Hexocyclium methylsulfate	freely soluble	
	Homatropine methylbromide	1 in 6	9.9
	Hyoscyamine sulphate	2 in 1	3.5
	Methantheline bromide	1 in 5	
10	Hyoscine hydrobromide	1 in 3	7.6
	Oxyphenonium bromide	freely soluble	3.2
	Propantheline bromide	very soluble	9.0
	Tridihexethyl chloride	1 in 3	
15	<u>Sympathomimetics</u>		
	Bitolterol Mesylate		9.1
	Ephedrine	1 in 20	9.6
	Ephedrine HCl	1 in 3	9.6
20	Ephedrine sulphate	1 in 1	9.6
	Orciprenaline sulphate	1 in 2	9.0, 10.1, 11.4
	Phenylpropanolamine		
	hydrochloride	1 in 2.5	9
25	Pseudoephedrine hydrochloride	1 in 1	9.8
	Ritodrine hydrochloride	1 in 10	9
	Salbutamol sulphate	1 in 4	9.3, 10.3
	Terbutaline sulphate	1 in 4	8.7, 10.0, 11.0
30	<u>Sympatholytic Agents</u>		
	Phenoxybenzamine		
	hydrochloride	1 in 25	4.4
	<u>Miscellaneous Autonomic Drugs</u>		
35	Nicotine	soluble	7.9
	<u>Iron Preparations</u>		
	Ferrous gluconate	1 in 10	
	Ferrous sulphate	1 in 5	
40	<u>Haemostatics</u>		
	Aminocaproic acid	1 in 1.5	4.43, 10.73
	<u>Cardiac Drugs</u>		
	Acebutolol HCl	1 in 5	9.4
45	Diltiazem hydrochloride	freely soluble	7.7
	Disopyramide phosphate	1 in 20	8.4
	Flecainide acetate	1 in 20	9.3
	Procainamide hydrochloride	1 in 0.25	9.23
50	Propranolol hydrochloride	1 in 20	9.5

	Quinidine Gluconate	freely soluble	4.0, 8.6
	Timolol maleate	freely soluble	9
5	Tocainide hydrochloride	freely soluble	7.8
	Verapamil hydrochloride	1 in 20	4-6.5
	<u>Antihypertensive Agents</u>		
	Captopril	freely soluble	3.7, 9.8
10	Clonidine hydrochloride	1 in 13	8.2
	Hydralazine hydrochloride	1 in 25	7.3
	Mecamylamine hydrochloride	1 in 5	11.2
	Metoprolol tartrate	very soluble	9.68
15	<u>Vasodilators</u>		
	Papaverine hydrochloride	1 in 2	6.4
	<u>Non-Steroidal Antiinflammatory Agents</u>		
	Choline salicylate	very soluble	
20	Magnesium salicylate	1 in 13	
	Meclofenamate sodium	freely soluble	4.0
	Naproxen sodium	soluble	4.15
	Tolmetin sodium	freely soluble	3.5
25	<u>Opiate Agonists</u>		
	Codeine HCl	1 in 30	8.2
	Codeine phosphate	1 in 4	8.2
	Codeine sulphate	1 in 30	8.2
30	Dextromoramide tartrate	1 in 25	7.1
	Hydrocodone bitartrate	1 in 10	8.3
	Hydromorphone hydrochloride	1 in 3	8.2
	Pethidine hydrochloride	very soluble	8.7
35	Methadone hydrochloride	1 in 2	8.3
	Morphine sulphate	1 in 21	8.0, 9.9
	Propoxyphene hydrochloride	1 in 0.3	
	<u>Anticonvulsants</u>		
40	Phenobarbital sodium	1 in 3	7.41
	Phenytoin sodium	soluble	8.3
	Troxidone	1 in 13	
	Ethosuximide	1 in 4.5	9.0
45	Valproate sodium	1 in 5	4.8
	<u>Tranquilizers</u>		
	Acetophenazine maleate	1 in 10	
	Chlorpromazine hydrochloride	1 in 0.4	9.3
50	Fluphenazine hydrochloride	1 in 10	3.9, 8.1
	Prochlorperazine edisylate	1 in 2	3.7, 8.1

	Promazine hydrochloride	1 in 1	9.4
	Thioridazine hydrochloride	1 in 9	9.5
5	Trifluoroperazine		
	hydrochloride	1 in 2	8.1
	Lithium citrate	1 in 2	
	Molindone hydrochloride	freely soluble	6.9
10	Thiothixine hydrochloride	1 in 8	
	<u>Stimulants</u>		
	Benzphetamine hydrochloride	freely soluble	6.6
	Dextroamphetamine sulphate	1 in 10	9.9
15	Dextroamphetamine phosphate	1 in 20	9.9
	Diethylpropion hydrochloride	freely soluble	
	Fenfluramine hydrochloride	1 in 20	9.1
	Methamphetamine hydrochloride	1 in 2	
20	Methylphenidate hydrochloride	freely soluble	8.8
	Phendimetrazine tartrate	freely soluble	7.6
	Phenmetrazine hydrochloride	1 in 0.4	8.4
	Caffeine citrate	1 in 4	14
25	<u>Barbiturates</u>		
	Amylobarbitone sodium	1 in 1	7.8
	Butabarbital sodium	freely soluble	7.9
	Secobarbital sodium	1 in 3	7.5
30	<u>Sedatives</u>		
	Hydroxyzine hydrochloride	1 in 1	2.6, 7.0
	Methyprylon	1 in 14	12
	<u>Expectorants</u>		
35	Potassium Iodide	1 in 0.7	
	<u>Antiemetics</u>		
	Benzaquinamide hydrochloride	1 in 10	5.9
40	Metoclopramide HCl	1 in 0.7	7.3, 9.0
	Trimethobenzamide		
	hydrochloride	1 in 2	8.3
	<u>GI Drugs</u>		
45	Ranitidine hydrochloride	1 in 2	8.2, 2.7
	<u>Heavy Metal Antagonists</u>		
	Penicillamine	1 in 9	1.8
	Penicillamine HCl	1 in 1	8.0, 10.8
50	<u>Antithyroid Agents</u>		
	Methimazole	1 in 5	

Genitourinary Smooth Muscle Relaxants

Flavoxate hydrochloride	freely soluble	
Oxybutynin hydrochloride	freely soluble	6.96

Vitamins

Thiamine hydrochloride	1 in 1	4.8, 9.0
Ascorbic acid	1 in 3	4.2, 11.6

Unclassified Agents

Amantadine hydrochloride	1 in 2.5	10.4
Colchicine	1 in 20	1.7, 12.4
Etidronate disodium	freely soluble	
Leucovorin calcium	very soluble	3.1, 4.8
		10.4
Methylene blue	1 in 25	-1
Potassium chloride	1 in 3	
Pralidoxime chloride	1 in 2	8

In the following description the active ingredient of high water solubility will be illustrated by reference to the opioid drug, morphine. However, this is illustrative only and the invention is in no way restricted thereto. Preferably, the active ingredient is an opiate selected from the group consisting of the salts of codeine, dextromoramide, hydrocodone, hydromorphone, pethidine, methadone, morphine and propoxyphene.

Morphine acts as an agonist primarily at mu, kappa and perhaps delta receptors in the central nervous system. By acting on these receptors the following pharmacological effects are seen. Analgesia due to a central action on pain perception, together with a modulatory effect on the central transmission of noxious sensation. It also causes drowsiness and euphoria (though sometimes dysphoria, particularly in those who are free of pain).

The pharmaceutical pellet composition according to the present invention may include a plurality of coated core elements.

The pharmaceutical composition may be provided in any suitable unit dosage form. An encapsulated form may be used.

The pharmaceutical pellet composition may be provided in a pellet or tableted pellet form. A tablet may be formed by compression of the pellets optionally with the addition of suitable excipients.

In a preferred aspect of the present invention the core coating, in use, generates a dissolution profile for the sustained release composition, which is equal to or greater than the minimum dissolution profile required to provide substantially equivalent bioavailability to a capsule, tablet or liquid containing an equal amount of the at last one active ingredient in an immediately available form.

"Dissolution profile" as used herein, means a plot of amount of active ingredient released as a function of time. The dissolution profile may be measured utilising the Drug Release Test (724) which incorporates standard test USPXXII 1990. (Test(711)). A profile is characterised by the test conditions selected. Thus the dissolution profile may be generated at a preselected shaft speed, temperature and pH of the dissolution media.

A first dissolution profile may be measured at a pH level approximating that of the stomach. At least a second dissolution profile may be measured at pH levels approximating that of at least one point in the intestine.

A highly acidic pH may simulate the stomach and a less acidic to basic pH may simulate the intestine. By the term "highly acidic pH" as used herein we mean a pH in the range of approximately 1 to 4. By the term "less acidic to basic pH" we mean a pH of greater than 4 up to approximately 7.5, preferably approximately 6 to 7.5.

A pH of approximately 1.2 may be used to simulate the pH of the stomach.

A pH of approximately 6.0 to 7.5 preferably 7.5 may be used to simulate the pH of the intestine.

Accordingly in a further preferred aspect, a first dissolution profile is measured at a pH level approximating that of the stomach and a second dissolution profile is measured at a pH level approximating that of at least one point in the intestine; the first and second dissolution profiles for the sustained release composition each being equal to or greater than the minimum dissolution required to provide substantially equivalent bioavailability to a capsule, tablet or liquid containing the at least one active ingredient in an immediately available form.

More preferably, the composition, in use, exhibits less fluctuations in plasma concentrations in active ingredient at steady state over a 24 hour period, relative to the active ingredient in an uncoated form and/or exhibits less diurnal variation in plasma concentration of active ingredient relative to known capsules or tablets containing the at least one active ingredient in a sustained release form.

For example, dissolution profiles have been generated which exhibit bioavailability substantially equivalent to, or better than, commercially known morphine compositions including MS Contin, MST Continus and morphine solution.

Accordingly, in a preferred aspect of the present invention there is provided a sustained release pharmaceutical pellet composition including

a core element including a morphine compound; and

a core coating for the core element which is partially soluble at a highly acidic pH to provide a slow rate of release of morphine compound and wherein the morphine compound is available for absorption at a relatively constant faster rate in the intestine over an extended period of time.

It will be understood that further since the active ingredient is provided in a sustained release pellet form significantly less fluctuations in plasma concentrations of active ingredients at steady state over a 24 hour period are encountered, and may allow for less frequent dosing relative to the active ingredient in an uncoated form. This is expected to result in less toxic and more effective therapeutic activity.

Similarly, it has been found that the pharmaceutical pellet composition according to the present invention exhibits less diurnal variation in plasma concentrations of active ingredient than prior art preparations, for example when administered on a two, three or four times daily dosage regimen.

Moreover, the pharmaceutical pellet composition according to the present invention shows no evidence of dose dumping. The relative bioavailability of the active ingredient generated from the pharmaceutical pellet composition is not compromised by food so that compliance will improve as the product may be taken without regard to meals.

Moreover, since the core coating is partially soluble at an acidic pH, for example as encountered in the stomach of the patients, some slow release of active ingredient will occur in the stomach. The slow rate of release of active ingredient may also be at a relatively constant rate.

The active ingredient may be available for absorption even in regions of the gastrointestinal tract which are not sufficiently alkaline to dissolve the enteric core coating component.

Thus the active ingredient is available for absorption in an absorption region substantially immediately after the pyloric sphincter in the patient. Such an absorption region may generally be characterised by a pH between approximately 1.2 and 5.5. Absorption will occur in the small intestine but since absorption will continue over an extended period of time, thus some absorption will occur additionally some way into the large intestine.

Where the active ingredient of high solubility in water is a morphine compound, the morphine compound may take any suitable form. The morphine compound may be present in an anhydrous or hydrous form. The morphine compound may be provided in a salt form. Morphine sulphate is preferred. Morphine sulphate pentahydrate is particularly preferred.

Advantages of the sustained release pharmaceutical pellet composition according to the present invention may thus be summarized as follows

- (i) The time during which morphine blood levels at steady state are greater than or equivalent to 75% of maximum blood levels ($t > 0.75C_{max}$) may be approximately 3 hours or greater. Generally $t > 0.75C_{max}$ may be 3.5 hours or greater ($t > 0.75C_{max}$ for MS Contin has been reported to be 3.5 hours).
- (ii) peak to trough variations in blood morphine concentrations at steady state will be between 60-100% (these variations for MS Contin have been reported to be approximately 300% and for Morphine Solution 4 hourly are approximately 200%)
- (iii) diurnal variations may be reduced
- (iv) the co-administration of food will not significantly decrease the extent of morphine absorption or alter the rate of morphine absorption when compared with administration in the fasted state the effect of food on morphine absorption from MS Contin is not known)
- (v) inter- and intra-subject variation in blood morphine pharmacokinetics may be reduced.

Accordingly, in a preferred aspect according to the present invention there is provided a sustained release pharmaceutical pellet composition including

a core element including at least one active ingredient of high solubility; and

a hybrid core coating which coating provides a slow rate of release of active ingredient at a highly acidic pH and a relatively constant faster rate of release at a less acidic to basic pH over an extended period of time.

Desirably, for some applications of the invention, the rate of release at the less acidic to basic pH is greater than the rate of release at the highly acidic pH, preferably 1.2 to three times greater.

The hybrid core coating may include

at least one polymer which is substantially insoluble independent of pH (insoluble matrix polymer); and at least one enteric polymer which is substantially insoluble at acidic pH but at least partially soluble at a less acidic to basic pH (enteric polymer);

at least one component which is at least partially soluble at acidic pH (acid soluble polymer).

It has been found necessary in order to achieve a slow rate of release at acidic pH for pH dependent or independent drugs, and faster relatively constant rate of release over an extended period of time to include the above three components in the hybrid core coating composition.

Preferably the enteric polymer is readily soluble at a less acidic to basic pH.

Preferably the at least partially soluble component is a readily water-soluble component.

Accordingly the hybrid core coating may include an effective amount of

a matrix (insoluble) polymer which is substantially insoluble independent of pH an enteric polymer whose solubility is pH dependent, and an at least partially acid soluble component.

The rate of dissolution at highly acidic pH of the hybrid core coating will depend on the amount of the at least one partially acid soluble component, the pH dependent and pH independent polymers, and the thickness of the coating. Typical core coatings may be in the range of approximately 5 to 200 μm , preferably approximately 25 to 50 μm . It will be understood, accordingly, that the rate of absorption may be modified by modifying the thickness and/or the composition of the hybrid core coating.

Once a minimum amount of the at least partially acid soluble component and/or the maximum thickness of the coating to achieve the minimum dissolution profile at an highly acidic pH has been established, then it is simply a matter of design choice to adjust the composition and/or thickness of coating as desired.

It has been found that the dissolution rate of the soluble drug at various pH's can be modified at will by altering the ratio of polymers. The ternary system of polymers according to the present invention allows greater flexibility than as known in prior art using only binary systems of polymers.

The at least one enteric polymer may be selected from cellulose acetate phthalate, hydroxypropyl methylcellulose phthalate (HPMCP), polyvinyl acetate phthalate, methacrylic acid copolymer, hydroxypropyl methylcellulose acetate succinate, shellac, cellulose acetate trimellitate and mixtures thereof. Particularly preferred enteric polymers include synthetic resin bearing carboxyl groups. The methacrylic acid: acrylic acid ethylester 1:1 copolymer sold under the trade designation "Eudragit L100-55" has been found to be suitable.

The at least one enteric polymer may be present in the coating in an amount of from approximately 1 to 60% by weight, preferably 2 to 20% by weight, more preferably 5 to 15% by weight, based on the total weight of the hybrid core coating excluding weight of filler and plasticiser.

The at least partially acid-soluble component may be selected from polymers such as polyvinyl pyrrolidone, hydroxypropyl cellulose, hydroxypropyl methylcellulose, polyethylene glycol, polyvinyl alcohol and monomers therefor such as sugars, salts, or organic acids and mixtures thereof.

The at least partially acid-soluble component may be present in the coating in amounts of from approximately 1 to 60%, preferably 15 to 40% by weight, more preferably 20 to 35% by weight, based on the total weight of the hybrid core coating excluding weight of filler and plasticiser.

The at least one insoluble matrix polymer may be any suitable pharmaceutically acceptable polymer substantially insoluble independent of pH. The polymer may be selected from ethylcellulose, acrylic and/or methacrylic ester polymers or mixtures thereof and the like may be used. Polymers or copolymers of acrylates or methacrylates having a low quaternary ammonium content may be used. The acrylic acid ethyl ester: methacrylic acid methyl ester (1:1) copolymer has been found to be suitable.

The at least one insoluble matrix polymer may be present in the coating in an amount of from approximately 1 to 85% by weight preferably 35 to 75% by weight, more preferably 45 to 65% by weight, based on the total weight of the hybrid core coating excluding weight of filler and plasticiser.

The hybrid core coating may further include at least one plasticiser; and optionally at least one filler.

Accordingly in a preferred aspect the hybrid core coating includes

0 to approximately 50% by weight, preferably 2.5 to 30% by weight, based on the total weight of the hybrid core coating of at least one plasticiser selected from diethyl phthalate, triethyl citrate, triethyl acetyl citrate, triacetin, tributyl citrate, polyethylene glycol and glycerol and the like; and

5 0 to approximately 75% by weight based on the total weight of the hybrid core coating of a filler selected from insoluble materials such as silicon dioxide, titanium dioxide, talc, alumina, starch, kaolin, polacrilin potassium, powdered cellulose, and microcrystalline cellulose and mixtures thereof.

The at least one plasticiser may be selected from diethyl phthalate, triethyl citrate, triethyl acetyl citrate, triacetin, tributyl citrate, polyethylene glycol and glycerol and the like. It will be understood that the plasticiser used may be largely dictated by the polymer used in the coating formulation, and the compatibility of the plasticiser with coating solution or dispersion. It should be noted that acid or water soluble plasticisers can also be used to function as the partially acid soluble component. The plasticiser may function to improve the physical stability of the core coating. A plasticiser is particularly preferred where the polymer has a high glass transition temperature and/or is of a relatively low molecular weight.

15 The plasticiser may be present in any suitable effective amount. Amounts of from approximately 0 to 50% by weight preferably 2.5 to 30% by weight based on the total weight of the hybrid core coating, have been found to be suitable.

The filler may be present in any suitable effective amount. Amounts of from 0 to approximately 75% by weight, preferably 15 to 60% by weight, more preferably 25 to 45% by weight, based on the total weight of the hybrid core coating have been found to be suitable.

Accordingly in a further preferred aspect the hybrid core coating has a formulation

	Ethylcellulose	45 to 60%)	% excluding
25	Methacrylic acid)	'plasticiser
	acrylic acid ethyl ester)	and filler
	1:1 copolymer	5 to 15%)	
	Polyethylene glycol	20 to 35%)	
30	Diethyl phthalate	2.5 to 30%	
	Talc	25 to 45% of total weight	
		of hybrid core coating	

35 In a preferred aspect of the present invention the core element of the pharmaceutical composition according to the present invention may include an effective amount of

at least one active ingredient of high solubility;

at least one core seed; and

40 at least one binding agent.

The active ingredient may be present in any suitable effective amount. The amount of active ingredient is dependent on the potency of the active ingredient and on the desired dosage strength and volume of a unit dose of the drug product. The active ingredient may be present in amounts of approximately 0.1 to 95% by weight, based on the total weight of the core element. The active ingredient may preferably be a morphine compound. The morphine compound may be present in amounts of approximately 10 to 60% by weight, based on the total weight of the core element.

The binding agent may be present in amounts of from approximately 0.1 to 45% by weight preferably approximately 0.1 to 20% by weight based on the total weight of the core element.

The binding agent may be of any suitable type. Suitable binders may be selected from polyvinyl pyrrolidone, hydroxypropyl cellulose, hydroxypropyl methylcellulose, methylcellulose and hydroxyethyl cellulose, sugars and mixtures thereof. The binding agent may be provided in the form of a granulating solution. An aqueous or organic solvent may be included. Methanol, ethanol or mixtures thereof may be used as solvents.

50 The size and amount of the core seed may vary substantially from approximately 100 μ m to 1700 μ m depending upon the amount of active ingredient to be included. Accordingly, the core seeds may vary from approximately 5 to 99% by weight, preferably 40 to 90% by weight based on the total weight of the core element, depending on the potency of the active ingredient. The core seed may be of such a diameter to provide a final core element having a diameter of approximately 200 to 2000 μ m.

The core seed may be of any suitable type. A sugar or an active core seed may be used.

The core element may further include other carriers or excipients, fillers, stabilizing agents and colorants. Suitable fillers may be selected from insoluble materials such as silicon dioxide, talc, titanium dioxide, alumina, starch, kaolin, polacrilin potassium, powdered cellulose, and microcrystalline cellulose and mixtures thereof. Soluble fillers may be selected from mannitol, sucrose, lactose, dextrose, sodium chloride, sorbitol and mixtures thereof.

In a preferred aspect the core element includes an effective amount of at least one morphine compound; optionally
at least one core seed; and
at least one binding agent.

The core element may have a formulation

Morphine sulphate	10 to 60% by weight
Core seeds	30 to 89.9% by weight
Hydroxypropylmethylcellulose	0.1 to 10% by weight

Alternatively the core element may have a formulation

Morphine sulphate	10 to 60% by weight
Core seeds	30 to 87.5% by weight
Polyvinyl pyrrolidone	2.5 to 10% by weight

The hybrid core coating composition may be provided in the form of a solution, dispersion or suspension.

In the form of a solution, the solvent may be present in amounts of from approximately 25 to 97% by weight, preferably 85 - 97%, based on the total weight of the hybrid core coating composition. The solvent for the polymer may be a solvent such as water, methanol, ethanol, methylene chloride and mixtures thereof.

In the form of a dispersion or suspension, the diluting medium may be present in amounts of from approximately 25 to 97% by weight, preferably 75 - 97%, based on the total weight of the hybrid core coating composition and is comprised predominantly of water.

Typical hybrid core coating formulations may be prepared in the amounts as follows:

Core Coating Formulation

A. Insoluble matrix polymer				45	- 65%)	% excluding
Enteric				4	- 10%)	solvent and filler
Acid soluble				15	- 35%)	
Plasticiser				4	- 30%)	
Solvent				85	- 97% of total coating	
						formula.
B. Insoluble matrix polymer				45	- 65%)	% excluding
Enteric				4	- 15%)	solvent and filler
Acid Soluble				15	- 35%)	
Plasticiser				4	- 30%)	
Diluting medium				75	- 97% of total coating	
						formula

Optionally, an amount of filler not exceeding 50% of the core coating formulations weight excluding solvent, may be added.

In a further aspect of the present invention, there is provided a method for preparing a sustained release pharmaceutical pellet composition, which method includes

- 5 providing
 - a core element including
 - at least one active ingredient of high solubility; and
 - at least one binding agent; and
 - a hybrid core coating composition including a solution, suspension or dispersion of
 10 at least one polymer which is substantially insoluble independent of pH;
 - at least one enteric polymer which is substantially insoluble at acidic pH but at least partially soluble at a less acidic to basic pH; and
 - at least one component which is at least partially soluble at acidic pH;
 - introducing the core element into a fluidised bed reactor; and
 - 15 spraying the hybrid core coating composition onto the core element.

In a preferred aspect the method may further include the preliminary steps of providing

- at least one active ingredient of high solubility;
- at least one binding agent;
- 20 at least one core seed; and
- coating the core seeds with the active ingredient and binding agent to form a core element.

In an alternative form the at least one binding agent is provided in a granulating solution. In this form the coating step may be conducted as a spheronisation process. The spheronisation process includes contacting the core seeds with the active ingredient and simultaneously adding the granulating solution thereto. The spheronisation process may be conducted in a spheronising machine.

25 In a further alternative aspect of the present invention, the method may further include the preliminary steps of

- providing
 - at least one active ingredient of high solubility;
 - 30 at least one binding agent; and
 - an effective amount of a solvent,
 - mixing the ingredients; and
 - subjecting the ingredients to an extrusion followed by marumerisation to form a core element.

The solvent may be an aqueous or organic solvent or mixtures thereof. The solvent may be present in 35 an amount effective to allow the ingredients to be extruded.

The core elements formed are then subjected to a drying step. The drying step may be conducted in a fluidised bed or drying oven.

In a preferred form the at least one binding agent and active ingredient are provided in a solution or slurry. In this form the core seeds are sprayed with the solution or slurry. The spraying step may be 40 conducted in any suitable coating equipment. The coating equipment may be a fluidised bed chamber, preferably a rotary fluid bed machine.

Spray coating of core elements may be undertaken utilising bottom, top or tangentially located spray nozzles. A bottom spray nozzle may reside proximate to the base of the fluidised bed facing upwards while a top spraying nozzle is located above the contents of the bed and facing downwards. The spray nozzle 45 may reside in the mid-section of the fluidised bed and be oriented such as to spray tangentially to the rotating core elements.

The sustained release pharmaceutical pellet composition may be administered under a similar dosage regimen to that used in the prior art. The multi-pellet encapsulated form may for example be administered every eight to twenty-four hours in sustained release form.

50 In a further preferred aspect of the present invention the pharmaceutical pellet composition incorporating morphine compound may provide effective pain relief with twice or three times or four times daily administration. Versatility of dosing may be achieved with 10 mg, 20 mg, 50 mg, 100 mg, 200 mg, 500 mg or any other dose strength of capsules required.

The pharmaceutical pellet composition may be in multi-pellet encapsulated, sprinkle sachet or tableted 55 forms.

In accordance with a further aspect of the present invention, there is provided a method of treating pain-associated conditions in patients requiring such treatment which method includes administering to a patient an effective amount of a sustained release pharmaceutical pellet composition including

a core element including at least one morphine compound of high solubility; and

a core coating for the core element which is partially soluble at a highly acidic pH and wherein the morphine compound is available for absorption at a relatively constant rate in the intestine over an extended period of time.

5 The method of treatment according to this aspect of the present invention is particularly applicable to the treatment of acute and chronic pain, particularly pain associated with terminal disease such as cancer and chronic backpain, as well as post-operative pain.

Preferably the pharmaceutical sustained release composition is provided in a unit dosage form and administration occurs at intervals of approximately 8 to 24 hours.

10 The present invention will now be more fully described with reference to the accompanying examples. It should be understood, however, that the following description is illustrative only and should not be taken in any way as a restriction on the generality of the invention specified above.

EXAMPLE 1

1. Formulation 1

Core Composition 1	
Morphine Sulphate pentahydrate	194 g
Core seeds	170 g
Polyvinyl pyrrolidone	37 g
Ethanol	185 g

Hybrid Core Coating Composition 1	
Polyethylene Glycol	12 g
Ethylcellulose	25 g
Diethyl phthalate	2 g
Methacrylic acid : acrylic acid ethyl ester 1:1 copolymer	5 g
Talc	22 g
Ethanol	667 g

2. Formulation 2

Core Composition 2	
Morphine Sulphate pentahydrate	194 g
Core Seeds	170 g
Polyvinyl pyrrolidone	37 g
Ethanol	185 g

Hybrid Core Coating Composition 2	
Polyethylene Glycol	25 g
Ethylcellulose	41 g
Diethyl phthalate	3 g
Methacrylic acid: acrylic acid ethyl ester 1:1 copolymer	4 g
Talc	37 g
Ethanol	1106 g

3. Formulation 3

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Core Composition 3	
Morphine Sulphate Pentahydrate	364 g
Core Seeds	733 g
Hydroxypropylmethylcellulose	14 g
Ethanol	986 g

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Hybrid Core Coating Composition 3	
Polyethylene Glycol	47 g
Ethylcellulose	90 g
Diethyl phthalate	19 g

20 Spheronised Core Manufacture (Core Composition 1 and 2)

The core seeds were placed in a spheroniser. The core seeds were then coated with a dry mixture of the active ingredients and inactive excipients whilst concomitantly adding a solution of the binder components.

25 The wet cores so formed were then dried in a fluidised bed dryer for 1 hour.

Rotacoating Core Manufacture (Core Composition 3)

30 The core seeds were placed in a rotor fluid bed machine. The core seeds were then coated with a suspension or solution of the active ingredients and inactive excipients including at least one binding agent, in a suitable liquid. The wet cores so formed were then dried in a suitable drier for one hour.

Pellet Manufacture

35 (a) The dried spheronised cores 1 and 2 were then placed in a fluid bed coating apparatus. The hybrid core coating compositions 1 and 2 were then sprayed onto the cores 1 and 2 to form Formulation 1 and 2 pellets respectively. At the conclusion of the process, the pellets were fluid bed dried.

(b) The dried cores 3 were then placed in a rotary fluid bed or conventional fluid bed coating apparatus. The hybrid core coating composition 3 was then sprayed onto the cores 3 to form Formulation 3 pellets.

40 A dissolution test was conducted on the pellet compositions 1, 2 and 3 utilizing the test method USPXXII 1990 (Test 711). A sample is dissolved in an aqueous medium previously degassed and equilibrated to 37°C. The media are USP pH 1.2 media without enzymes and pH 7.5 phosphate buffer. A sample of known volume is withdrawn at designated time intervals from the bath as directed and subjected to a suitable assay procedure. The mg of morphine sulphate as a function of time is plotted as the

45 dissolution profile.

The tests were conducted at pH 1.2 and pH 7.5.

The baskets containing the samples were rotated at approximately 50 r.p.m. and the aqueous medium maintained at approximately 37°C.

50 The results are given in Tables 1 to 6 and Figures 1 and 6 herein. The results for Formulation 1 at pH 1.2 and 7.5 are given in Tables 1 and 2 respectively. The hybrid coating on Formulation 1 pellet allows dissolution at pH 1.2, a significantly faster rate of dissolution is observed at pH 7.5. The results for Formulation 2 pellet at pH 1.2 and 7.5 are given in Tables 3 and 4 respectively, and are similar to those obtained from composition A.

55 The results for Formulation 3 pellets are similar to those achieved for Formulation 1 at pH 7.5. The results achieved for Formulation 3, however, illustrate a significant prolongation of release thereover.

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TABLE 1

DISSOLUTION DATA FOR FORMULATION 1 AT pH 1.2 (AVERAGED DATA FOR 3 SAMPLES)				
TIME MIN.	MG RELEASED	SD	% RELEASED	SD
30	0.00	0.00	0.00	0.00
60	2.29	0.09	4.04	0.15
120	8.43	0.18	14.88	0.28
180	14.66	0.39	25.87	0.71

TABLE 2

DISSOLUTION DATA FOR FORMULATION 1 AT pH 7.5 (AVERAGED DATA FOR 3 SAMPLES)				
TIME MIN.	MG RELEASED	SD	% RELEASED	SD
30	1.85	0.09	3.28	0.17
60	9.03	0.25	16.07	0.45
120	23.20	0.42	41.29	0.77
180	35.39	0.50	63.00	1.01

TABLE 3

DISSOLUTION DATA FOR FORMULATION 2 AT pH 1.2 (AVERAGED DATA FOR 3 SAMPLES)				
TIME MIN.	MG RELEASED	SD	% RELEASED	SD
30	1.64	0.00	3.22	0.01
60	6.26	0.09	12.25	0.16
120	20.24	0.18	39.63	0.46
180	36.39	0.27	71.27	0.72
240	47.47	0.49	92.97	1.12

TABLE 4

DISSOLUTION DATA FOR FORMULATION 2 AT pH 7.5 (AVERAGED DATA FOR 3 SAMPLES)				
TIME MIN.	MG RELEASED	SD	% RELEASED	SD
30	2.63	0.00	5.12	0.03
60	8.69	0.09	16.94	0.11
120	21.62	0.33	42.13	0.40
180	33.66	0.59	65.60	0.79
240	42.47	0.82	82.78	1.13

TABLE 5

DISSOLUTION DATA FOR FORMULATION 3 AT pH 1.2 (AVERAGED DATA FOR 3 SAMPLES)				
TIME MIN.	MG RELEASED	SD	% RELEASED	SD
30	1.44	0.39	2.12	0.53
60	3.03	0.33	4.48	0.39
120	6.78	0.30	10.03	0.36
180	10.17	0.18	15.04	0.34
240	13.87	0.41	20.51	0.29
300	17.45	0.31	25.81	0.30
360	21.29	0.21	31.49	0.27
420	24.75	0.32	36.62	0.46
480	28.60	0.64	42.30	0.37
540	32.63	0.42	48.28	0.45
600	35.80	0.92	52.95	0.37
24 hours	67.60	1.26	100.04	3.79

TABLE 6

DISSOLUTION DATA FOR FORMULATION 3 AT pH 7.5 (AVERAGED DATA FOR 3 SAMPLES)				
TIME MIN.	MG RELEASED	SD	% RELEASED	SD
30	2.19	0.11	3.23	0.17
60	7.05	0.89	10.38	1.26
120	18.07	1.05	26.63	1.44
180	28.12	1.03	41.44	1.35
240	37.86	1.05	55.80	1.32
300	47.60	1.48	70.16	1.96
360	56.33	0.54	83.03	0.47
420	63.03	2.01	92.90	2.76
480	65.97	0.61	97.23	0.75
540	69.13	0.41	101.89	0.79
600	70.20	0.43	103.47	0.45
24 hours	74.76	2.36	110.19	3.04
SD = Standard Deviation				

EXAMPLE 2

Two sustained release morphine compositions in accordance with the present invention have been trialled in patients with back pain (fed and fasting) and in healthy volunteers (fasting). The results of these trials suggest that Faulding already has a product that is superior to a commercial product MS Contin with regard to sustained delivery of morphine. An investigation has also been initiated into understanding the effect that food has on the absorption of morphine.

The sustained release oral morphine compositions according to the present invention are designated Formulation 1 and Formulation 2.

1. PART A

A single dose 3 way crossover study under fasted conditions was conducted in six patients suffering chronic pain. On 3 occasions separated by one week, patients received a 50 mg oral morphine dose as either a solution (reference dose) or one of two sustained release formulations as pellets contained within a capsule (designated Formulation 1, a pH dependent release formulation; and Formulation 2, a pH independent release formulation). The doses were administered after an overnight fast. Venous blood

samples were taken at specified time intervals from immediately after dose administration for 30 hours after the sustained release formulations and for 10 hours after the reference solution dose. The morphine concentration in the blood samples was quantitated using high pressure liquid chromatography (HPLC) with electrochemical detection. Table 3.1 summarizes the mean area under the curve (AUC); C_{max} (maximum blood concentration); T_{max} (time to reach peak blood concentration); $T_{1/2}$ (apparent terminal half life); $T_{\geq 0.75 C_{max}}$ (time for which blood concentration was greater than 75% of C_{max}) and relative bioavailability (F%).

The results revealed that both Formulation 1 and Formulation 2 provide a sustained release relative to the reference solution as assessed by:

- (1) a lower C_{max} for the formulations;
- (2) a longer T_{max} for the formulations; and
- (3) a longer time for which the blood morphine concentration was above 0.75 C_{max} for the formulations.

There was a significant decrease in C_{max} values for each formulation compared with the reference solution. The mean (\pm SD) C_{max} for the solution was 73.6 ± 30.9 ng/mL whereas the corresponding values for Formulation 1 and Formulation 2 were 21.6 ± 7.1 ng/mL and 23.2 ± 4.8 ng/mL respectively. The variability in C_{max} for Formulations 1 and 2 as demonstrated by the coefficient of variation was significantly less than that of the solution in the same patients.

There was a significant increase in T_{max} values for the formulations relative to that obtained with the reference solution. The mean (\pm SD) T_{max} for solution was 1.07 ± 1.09 hours whereas the equivalent values were 5.33 ± 1.2 hours and 4.25 ± 1.33 hours for Formulations 1 and 2 respectively. The variability in T_{max} values for the formulations was less than that obtained for the solution in the same patients.

The time the blood morphine concentration was greater than or equal to 0.75 C_{max} was significantly greater for the formulations compared to the reference solution dose. The mean time was 190 minutes for Formulation 1 and 237 minutes for Formulation 2 compared to only 59 minutes for the reference solution. Expressing these data as percentage of the time of the reference solution, Formulation 1 was 322% while Formulation 2 has 400% greater time that the blood morphine concentration was greater than 0.75 C_{max} compared to the solution.

There was no significant difference between the AUC for the formulations and that obtained for the reference solution (Table 3.1).

The relative bioavailability for the formulations was calculated from the ratio of the AUC for the appropriate formulation relative to that obtained for the reference solution for each patient. The relative bioavailability was 83.5% for Formulation 1 and 102.6% for Formulation 2.

The AUC and relative bioavailability data suggest that the extent of absorption of morphine from the three different formulations is similar whereas the C_{max} , T_{max} and $T_{\geq 0.75 C_{max}}$ data indicate that the formulations exhibit the typical slower and prolonged absorption of a true sustained release preparation.

TABLE 3.1

RESULT OF STUDY PART A					
PARAMETER	SOLUTION	FORMULATION 1		FORMULATION 2	
	MEAN	MEAN	OBSERVED DIFF	MEAN	OBSERVED DIFF
AUC (ng.h/mL)	199.77	170.72	-29.05	193.77	-6.0
SD	±66.32	±86.3		±46.35	
CV%	33	51		24	
C _{max} (ng/mL)	73.57	21.60	-52.0	23.16	-50.4
SD	±30.92	±7.12		±4.76	
CV%	42	33		21	
T _{max} (hours)	1.07	5.33	4.26	4.25	3.18
SD	±1.1	±1.21		±1.33	
CV%	103	23		31	
Bioavailability (F%)	100.0	83.53	-16.47	102.62	2.62
SD	±0.00	±27.87		±25.77	
CV%	0	33		25	
t _{1/2} (hours)	3.02	6.58	3.56	7.65	4.63
SD	±1.97	±5.33		±5.59	
CV%	65	81		73	
T _{≥0.75 C_{max}} (minutes)	59.0	189.8	130.8	237.3	178.3
SD	±37	±76		±95	
CV%	63	40		40	

2. PART B

A single dose 3 way crossover study under fed conditions was conducted in six patients suffering chronic pain. The same patients took part in both Parts A and B of this study. On 3 occasions separated by one week, patients received a 50 mg oral morphine dose as either a solution (reference dose) or one of two sustained release formulations as pellets contained within a capsule (designated Formulation 1, a pH dependent release formulation; and Formulation 2, a pH independent release formulation). The doses were administered after an overnight fast. Venous blood samples were taken at specified time intervals from immediately after dose administration for 30 hours after the sustained release formulations and for 10 hours after the reference solution dose. The morphine concentration in the blood samples was quantitated using high pressure liquid chromatography (HPLC) with electrochemical detection. Table 3.2 summarises the mean area under the curve (AUC); C_{max} (maximum blood concentration); T_{max} (time to reach peak blood concentration); T_{≥0.75 C_{max}} (time for which blood concentration was greater than 75% of C_{max}) and relative bioavailability (F%).

The results revealed that, in the presence of food, both Formulation 1 and 2 provide a sustained release relative to the reference solution as assessed by:

- (1) a lower C_{max} for the formulations;
- (2) a longer T_{max} for the formulations; and
- (3) a longer time for which the blood morphine concentration was above 0.75 C_{max} for the formulations.

There was a significant decrease in C_{max} values for each formulation compared with the reference solution. The mean (±SD) C_{max} for the solution was 80.7 ± 26.4 ng/mL whereas the corresponding values for Formulation 1 and Formulation 2 formulations were 22.0 ± 8.1 ng/mL and 32.6 ± 18.1 ng/mL respectively. The variability in C_{max} for Formulation 1 and 2 as demonstrated by the coefficient of variation was similar for all formulations. The C_{max} values for each formulation obtained under fed conditions were similar to the values obtained in the same patients under fasting conditions

(Part A).

There was a significant increase in T_{max} values for the formulations relative to that obtained with the reference solution. The mean (±SD) T_{max} for solution was 1.32 ± 1.65 hours whereas the equivalent values were 5.83 ± 0.75 and 4.5 ± 0.84 hours for Formulation 1 and 2 respectively. The variability in T_{max} values

for the formulations was less than that obtained for the solution. The T_{max} values were similar under fed and fasted conditions for each respective formulation.

The time the blood morphine concentration was greater than or equal to $0.75 C_{max}$ was significantly greater for the formulations compared to the reference solution dose. The mean time was 231.2 minutes for Formulation 1 and 168.5 minutes for Formulation 2 compared to only 52.2 minutes for the reference solution. Expressing these data as percentage of the time of the reference solution, Formulation 1 was 443% while Formulation 2 has 323% greater time that the blood morphine concentration was greater than $0.75 C_{max}$ compared to the solution. The data for the time greater than $0.75 C_{max}$ under fed and fasting conditions was similar for each respective formulation.

Under fed conditions, there was a significant difference between the AUC for the formulations and that obtained for the reference solution (Table 3.2) the reference solution having a greater AUC than either formulation. The mean areas were very similar for the formulations with mean values of 204.13 ± 106.11 ng.h/mL and 225.09 ± 138.52 ng.h/mL for Formulation 1 and Formulation 2 respectively. The mean AUC for the solution under fed conditions was 281.98 ± 112.58 ng.h/mL. The intersubject variability in AUC was similar for all formulations as shown by the coefficient of variation.

A comparison of AUC data obtained under fed and fasted conditions shows that the AUC for the reference solution expressed as a ratio of fed/fasted was 1.41 (range 0.94 to 1.9) with all but one patient having a ratio of greater than unity. There was a similar trend with the Formulations in that the mean AUC obtained when the formulations were administered immediately after food were larger than the equivalent value obtained in the fasted state.

The primary concern was to establish that "dose dumping" did not occur with either formulation. The data indicate that the bioavailability of morphine from formulations in the presence of food is at least equivalent to and possibly greater than the bioavailability from the same formulation in the fasted state and that the formulations behave in a similar manner to the solution with regard to the influence of food on the absorption of morphine.

The relative bioavailability for the formulations relative to that obtained for the reference solution was 79.4% for Formulation 1 and 78.2% for Formulation 2.

The AUC and relative bioavailability data suggest that the extent of absorption of morphine from the formulations is similar but slightly less than the solution in the fed state whereas the C_{max} , T_{max} and $T_{\geq 0.75 C_{max}}$ data indicate that the formulations exhibit the typical slower and prolonged absorption of a true sustained release preparation.

TABLE 3.2

RESULT OF STUDY PART B					
PARAMETER	SOLUTION	MOLLY 1		MOLLY 2	
	MEAN	MEAN	OBSERVED DIFF	MEAN	OBSERVED DIFF
AUC (ng.h/mL)	281.98	204.13	-77.85	225.09	-56.89
SD	± 112.58	± 106.11		± 138.52	
CV%	40	52		62	
C_{max} (ng/ml)	80.66	22.00	-58.66	32.63	-48.03
SD	± 26.44	± 8.05		± 18.07	
CV%	33	37		55	
T_{max} (hours)	1.32	5.83	4.51	4.50	3.18
SD	± 1.65	± 0.75		± 0.84	
CV%	125	13		19	
Bioavailability (F%)	100.0	79.4	-20.6	78.2	-21.8
SD	± 0.00	± 47.3		± 27.1	
CV%	0	60.0		35.0	
$T_{\geq 0.75 C_{max}}$ (minutes)	52.2	231.2	179.0	168.5	116.3
SD	± 39.3	± 73.9		± 55.5	
CV%	75	32		33	

Finally, it is to be understood that various other modifications and/or alterations may be made without departing from the spirit of the present invention as outlined herein.

Claims

1. A sustained release pharmaceutical pellet composition including
 - a core element including at least one active ingredient having an aqueous solubility of at least 1 in 30; and
 - a core coating for the core element which is partially soluble at a highly acidic pH, and wherein the active ingredient is available for absorption at a relatively constant rate in the intestine over an extended period of time.
2. A sustained release pharmaceutical pellet composition according to claim 1 wherein the active ingredient is an opiate agonist selected from the group consisting of the salt of codeine, dextromoramide, hydrocodone, hydromorphone, pethidine, methadone, morphine and propoxyphene.
3. A sustained release pharmaceutical pellet composition according to claim 2 wherein the active ingredient is a morphine compound.
4. A sustained release pharmaceutical pellet composition according to claim 2, wherein the core coating, in use, generates a dissolution profile as hereinbefore defined for the sustained release composition, which is equal to or greater than the minimum dissolution profile required to provide substantially equivalent bioavailability as hereinbefore defined to a capsule, tablet or liquid containing an equal amount of the active ingredient in an immediate available form.
5. A sustained release pharmaceutical pellet composition according to claim 4, wherein a first dissolution profile is measured at a pH level approximating that of the stomach and a second dissolution profile is measured at a pH level approximating that of at least one point in the intestine; the first and second dissolution profiles for the sustained release composition each being equal to or greater than the minimum dissolution required to provide substantially equivalent bioavailability to a capsule, tablet or liquid containing an equal amount of the active ingredient in an immediate available form.
6. A sustained release pharmaceutical pellet composition according to claim 5, wherein the composition, in use, exhibits less fluctuations in plasma concentrations of active ingredient at steady state over a 24 hour period, relative to the active ingredient in an uncoated form and/or exhibits less diurnal variation in plasma concentration of active ingredient relative to prior art capsules or tablets containing the active ingredient in a sustained release form.
7. A sustained release pharmaceutical pellet composition for administration to a patient at a predetermined dosage and interval which comprises a core element containing a therapeutically effective amount of at least one active ingredient having an aqueous solubility of at least 1 in 30 and a coating on said core element which comprises the following components:
 - (a) at least 35% by weight of a matrix polymer which is insoluble independent of pH;
 - (b) from 1 to 30% of an enteric polymer which is substantially insoluble at a pH of from 1 to 4, but which is soluble at a pH of from 6 to 7.5;
 - (c) from 1 to 60% of an acid-soluble compound soluble at a pH of from 1 to 4, sufficient to provide a slow rate of release of the active ingredient in the stomach;
 said percentages being weight based on the total weight of components (a), (b) and (c); the ratio of the components (a), (b) and (c) in said coating being effective such that the active ingredient is available for absorption at a relatively constant rate of release in the intestine such that the composition delivers to the patient a therapeutically effective amount of said active ingredient over the course of said predetermined interval, so as to maintain an active ingredient blood level at steady state of at least 75% of maximum blood level ($>0.75C_{max}$) for approximately 3.5 hours or greater and so that the time at which the active ingredient reaches its maximum concentration (t_{max}) is 4.5 hours or greater.
8. A sustained release pharmaceutical pellet composition according to claim 7 wherein the coating contains:
 - as component (a), ethyl cellulose, acrylic ester polymers, methacrylic ester polymers, an acrylic acid ethyl ester : methacrylic acid methylester (1:1) copolymer, or a mixture thereof;
 - as component (b), cellulose acetate phthalate, hydroxypropyl methylcellulose phthalate, polyvinyl acetate phthalate, methacrylic acid:acrylic acid ethylester 1:1 copolymer, hydroxypropyl methylcellulose

acetate succinate, shellac, cellulose acetate trimellitate and mixtures thereof; and
 as component (c), polyvinylpyrrolidone, hydroxypropyl cellulose, hydroxypropyl methylcellulose,
 polyethylene glycol, polyvinyl alcohol and monomers therefor and mixtures thereof.

- 5 9. A sustained release pharmaceutical pellet composition according to claim 8, wherein the coating
 comprises:
 35 to 75% by weight of component (a);
 2 to 20% by weight of component (b); and
 15 to 40% by weight of component (c).

10

10. A sustained release pharmaceutical pellet composition according to claim 9, wherein the coating also
 includes up to 50% of plasticizer selected from diethyl phthalate, triethyl citrate, triethyl acetyl citrate,
 triacetin, tributyl citrate, polyethylene glycol or glycerol and up to 75% of a filler selected from silicon
 dioxide, titanium dioxide, talc, alumina, starch, kaolin, polacrillin, potassium, powdered cellulose, and
 15 microcrystalline cellulose and mixtures thereof, said percentages being based on the total weight of the
 coating.

11. A sustained release pharmaceutical pellet composition according to claim 10, wherein the coating
 contains:

20

component (a)	35 to 75%
component (b)	2 to 20%
component (c)	15 to 40%
plasticizer	2.5 to 30%.

25

12. A sustained release pharmaceutical pellet composition according to any one of the preceding claims
 wherein the core element includes an effective amount of at least one active ingredient having an
 30 aqueous solubility of at least 1 in 30; at least one core seed; and at least one binding agent.

13. A sustained release pharmaceutical pellet composition according to claim 7, wherein the active
 ingredient is an opiate agonist selected from the group consisting of the salts of codeine, dex-
 tromoramide, hydrocodone, hydromorphone, pethidine, methadone, morphine and propoxyphene.

35

14. A sustained release pharmaceutical pellet composition according to claim 13 wherein the active
 ingredient is a morphine compound.

15. A method of treating pain-associated conditions in patients requiring such treatment which method
 40 includes administering to a patient an effective amount of a sustained release pharmaceutical pellet
 composition including

a core element including at least one morphine compound having an aqueous solubility of at least
 1 in 30; and

- 45 a core coating for the core element which is partially soluble at a highly acidic pH and wherein the
 morphine compound is available for absorption at a relatively constant rate in the intestine over an
 extended period of time, such that blood levels of morphine are maintained within the therapeutic range
 over an extended period of time

16. A method according to claim 15 wherein the pain associated conditions relate to the treatment of acute
 50 and chronic pain.

17. A method according to claim 16 wherein the sustained release pharmaceutical pellet composition is
 provided in a unit dosage form and administration occurs at intervals of approximately 8 to 24 hours.

55

FIGURE 1 - Morphine Sulphate Formulation 1
Dissolution Profile At pH 1.2 (Average Data For 3 Samples)

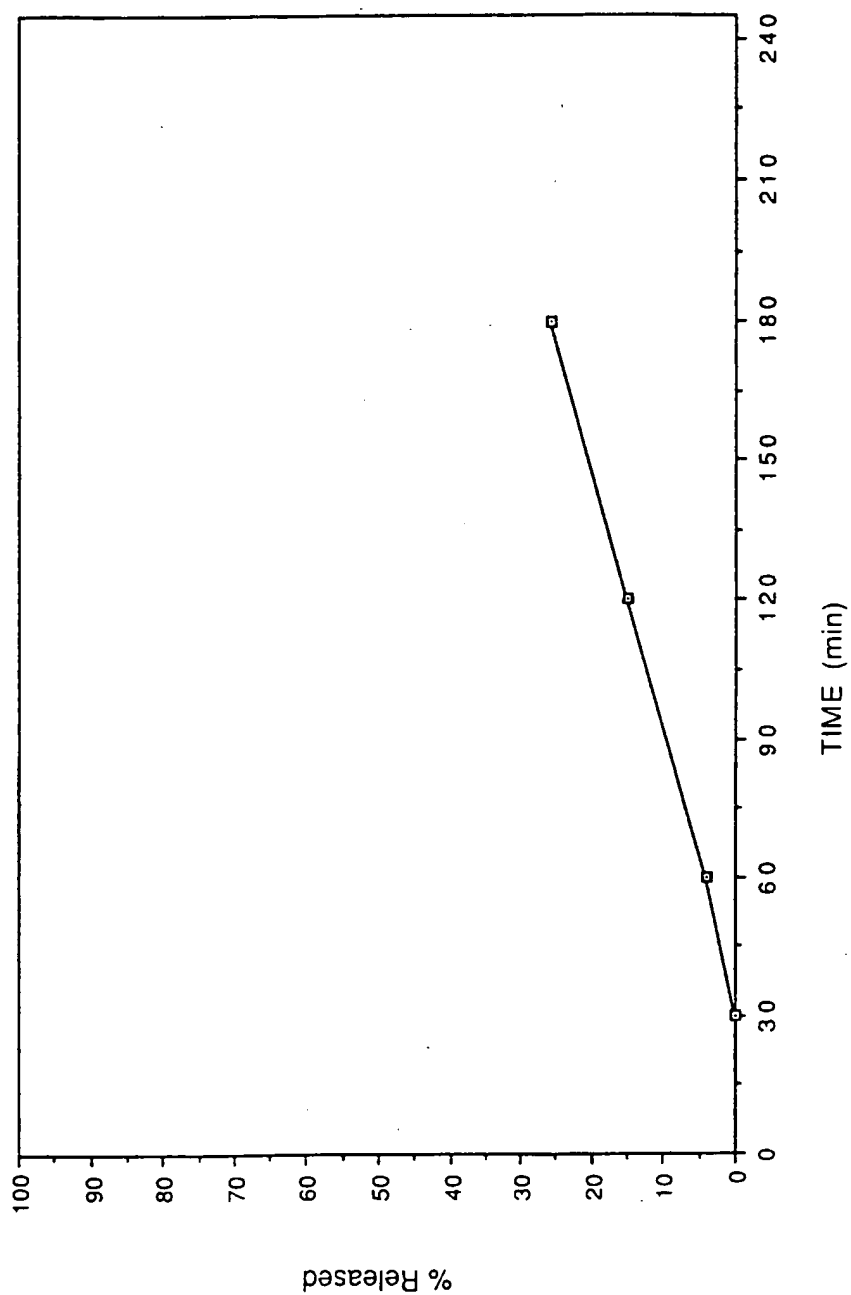


FIGURE 2 - Morphine Sulphate Formulation 1
Dissolution Profile At pH 7.5 (Average Data For 3 Samples)

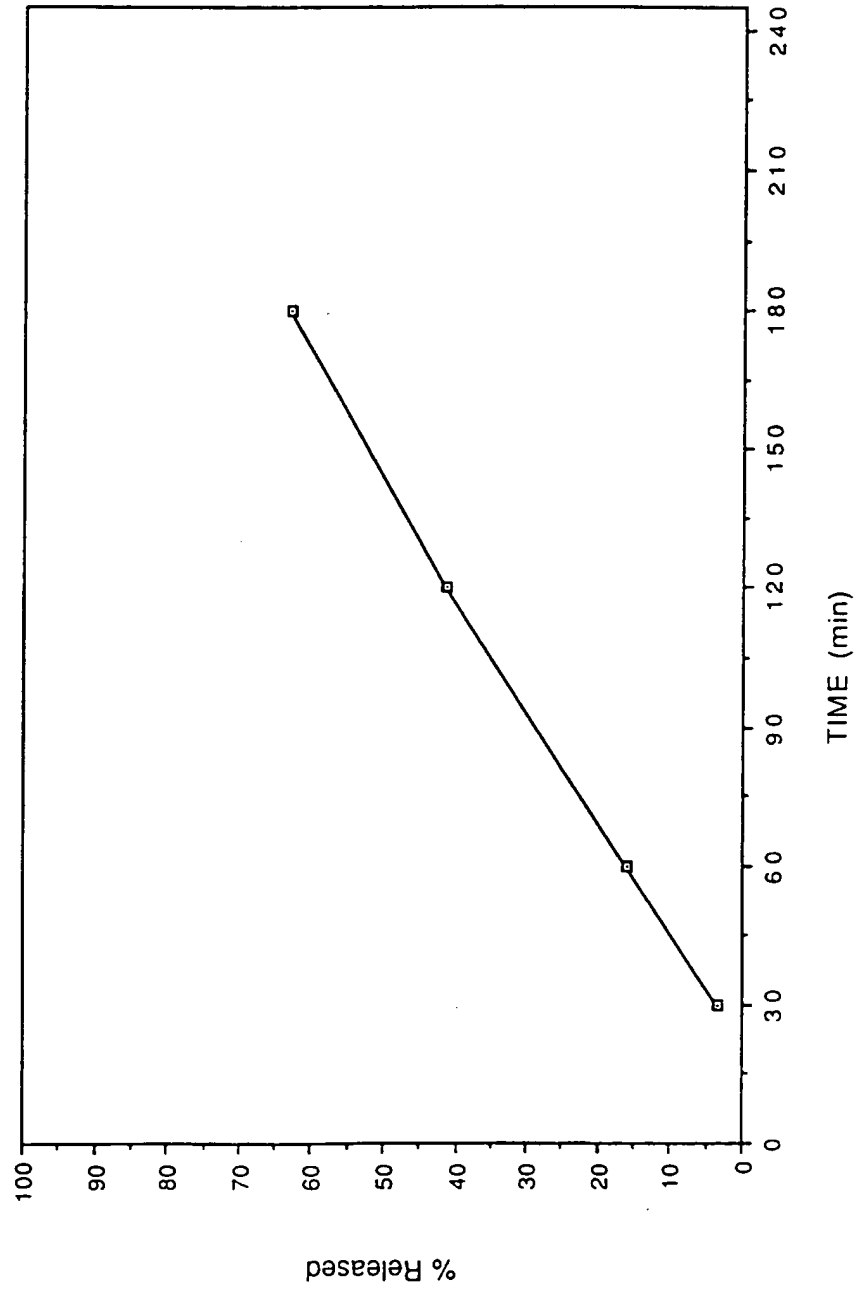


FIGURE 3 - Morphine Sulphate Formulation 2
Dissolution Profile At pH 1.2 (Average Data For 3 Samples)

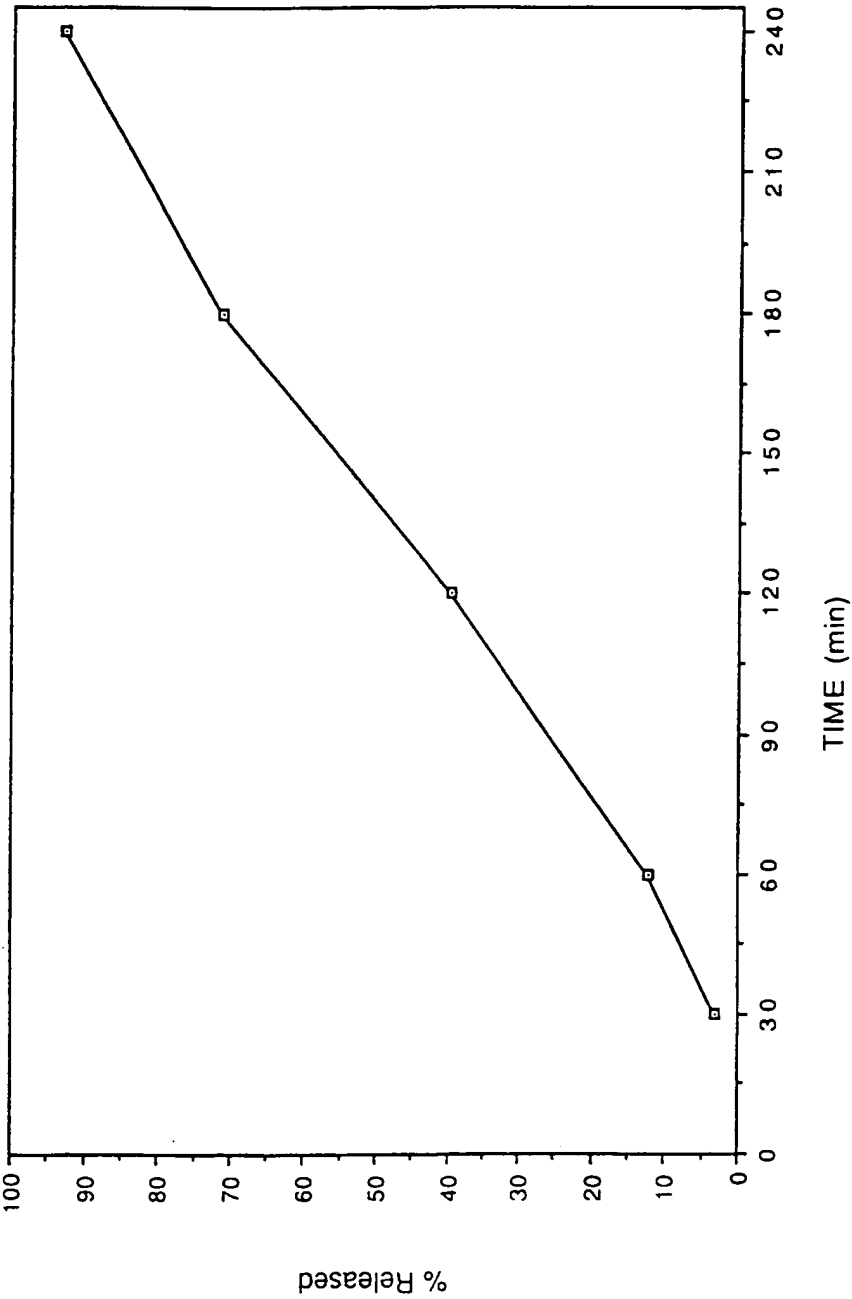


FIGURE 4 - Morphine Sulphate Formulation 2
Dissolution Profile At pH 7.5 (Average Data For 3 Samples)

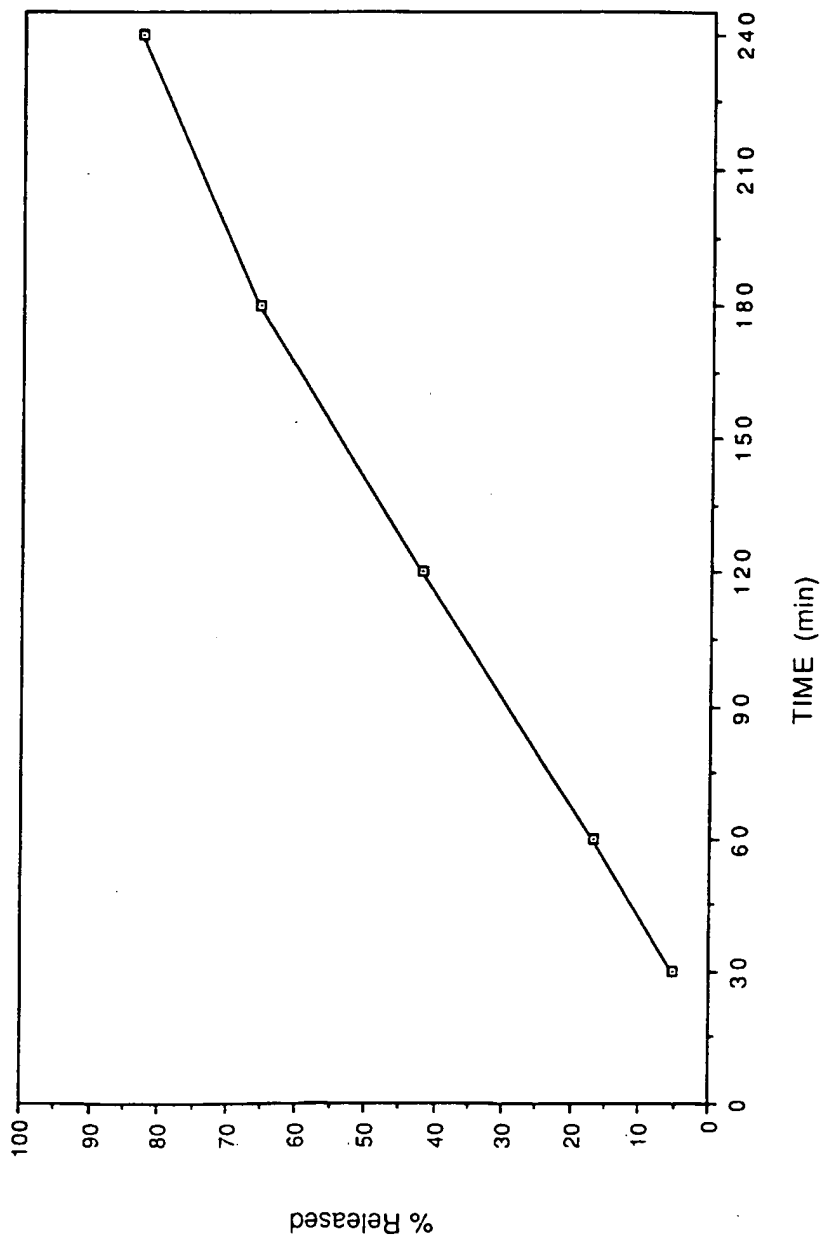


FIGURE 5 - Morphine Sulphate Formulation 3
Dissolution Profile At pH 1.2 (Average Data For 3 Samples)

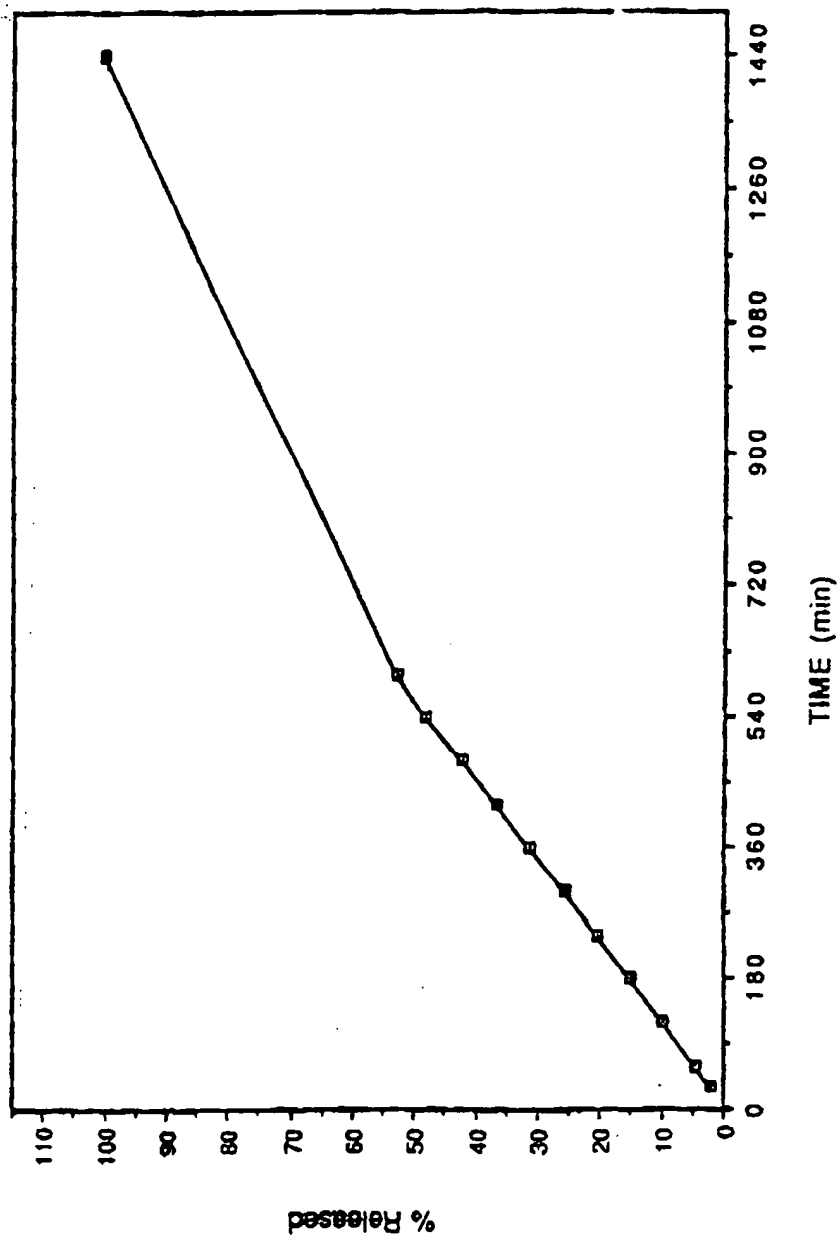
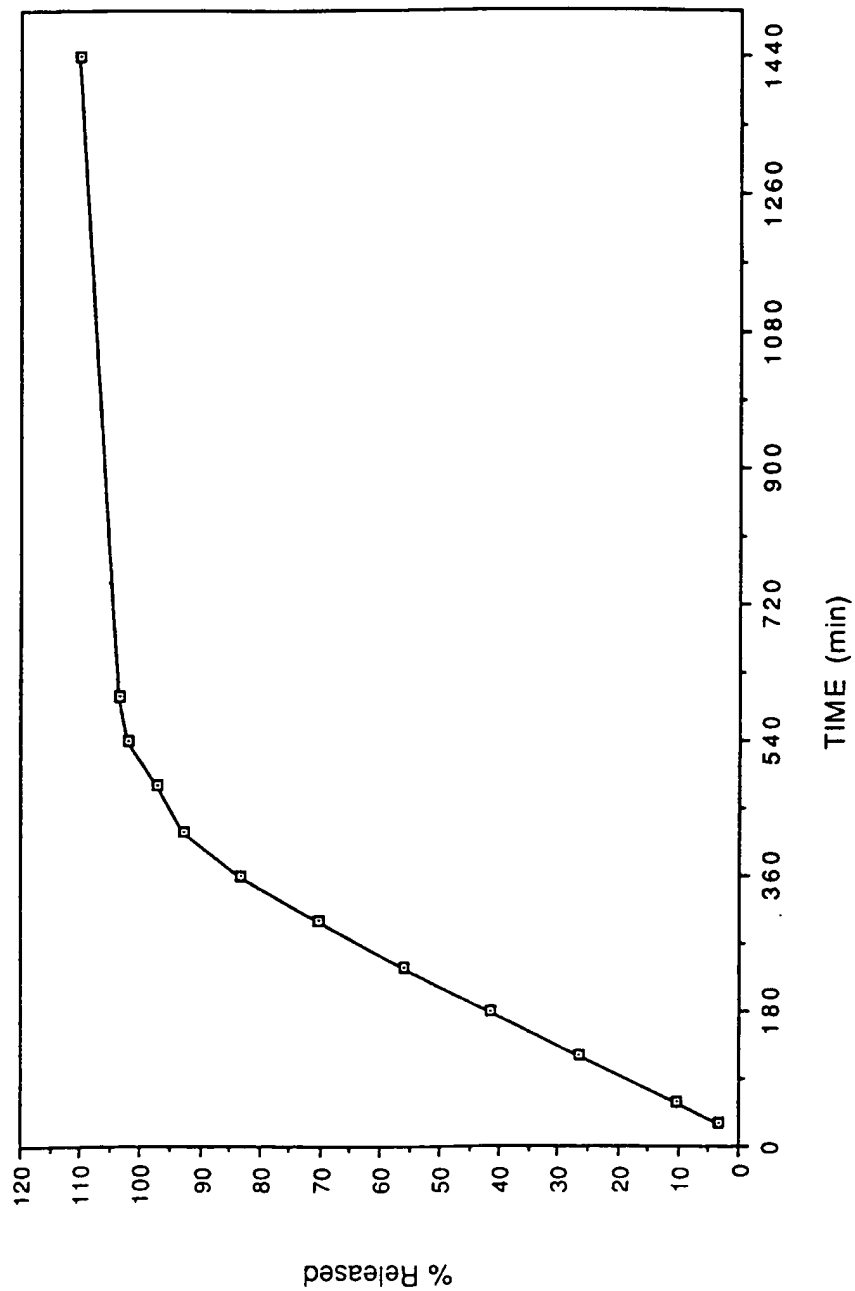


FIGURE 6 - Morphine Sulphate Formulation 3
Dissolution Profile At pH 7.5 (Average Data For 3 Samples)





European Patent
Office

PARTIAL EUROPEAN SEARCH REPORT

Application Number

which under Rule 45 of the European Patent Convention EP 94 20 0469 shall be considered, for the purposes of subsequent proceedings, as the European search report

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.5)
X Y	EP-A-0 260 236 (LEJUS MEDICAL AKTIEBOLAG) * column 1, line 5 - column 4, line 53 * * claims 1-7; figures 1-15B * ---	1,7-12 2-6, 13-17	A61K31/485 A61K9/52 A61K9/54
X Y	EP-A-0 164 959 (STERWIN AG) * page 3, line 9 - page 8, line 33 * * preparation A, example 1 * * page 12, line 26 - page 14, line 3 * * page 14, line 30 - page 15, line 24 * ---	1,7-12 2-6, 13-17	
X Y	EP-A-0 327 295 (F.H.FAULDING AND CO.LTD) * page 2, line 13 - page 3, line 15 * * page 3, line 35 - line 51 * * page 4, line 21 - page 5, line 7 * * page 6, line 2 - page 7, line 34 * --- -/--	1,7-12 2-6, 13-17	
			TECHNICAL FIELDS SEARCHED (Int.Cl.5)
			A61K
INCOMPLETE SEARCH			
<p>The Search Division considers that the present European patent application does not comply with the provisions of the European Patent Convention to such an extent that it is not possible to carry out a meaningful search into the state of the art on the basis of some of the claims</p> <p>Claims searched completely : Claims searched incompletely : Claims not searched : Reason for the limitation of the search:</p> <p>see sheet C</p>			
Place of search THE HAGUE		Date of completion of the search 20 May 1994	Examiner Sitch, W
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- & : member of the same patent family, corresponding document	

EP 0 FORM 1500 (04/92) (P.O. 0/07)



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PARTIAL EUROPEAN SEARCH REPORT

Application Number
EP 94 20 0469

DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int.Cls)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
Y	EP-A-0 205 282 (EUROCELTIQUE SA) * column 1, line 1 - line 5 * * column 3, line 29 - column 5, line 23; example 1 * * column 11, line 36 - column 12, line 36; figures 2,3 * -----	2-6, 13-17	
			TECHNICAL FIELDS SEARCHED (Int.Cl.5)



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Remark:

Although claims 15-17 are directed to a method of treatment of the human/animal body (Article 52(4) EPC) the search has been carried out and based on the alleged effects of the compound/composition.